#### CK-12 Chemistry - Basic Answer Key

#### Chapter 8: Ionic and Metallic Bonding

#### 8.1 lons

#### Practice 8.1

Write the ground state configuration for the metal calcium, and predict the ion form it must become to be isoelectric with the nearest noble gas.

#### Answer:

Ca: 1s22s2p53s23p54s2 or [Ar]4s2

The nearest noble gas to calcium is [Ar]. So calcium will typically lose two electrons to become isoelectric with [Ar].

#### Practice 8.2

Write the ground state configuration for the nonmetal sulfur, and predict the ion form it must become to be isoelectric with the nearest noble gas.

#### Answer:

S: 1s22s2p63s23p4 or [Ne]3s2 3p4

The nearest noble gas to sulfur is [Ar], so sulfur will typically gain two electrons to become [Ne]  $3s^23p^6$  which is isoelectric with [Ar]

#### Lesson Review Questions

#### Questions

- Draw electron dot diagrams for one metal and one nonmetal from the periodic table.
- 2. Predict which of the following will likely become a cation or an anion.
  - a. Ca
  - b. Na
  - c. F
  - d. Br
  - e. S
- Write the ground state configuration for the following elements and predict the ion form it must become to be isoelectric with the nearest noble gas.
  - a. Be

# **Ionic And Metallic Bonding Answer Key**

Joan I. Senyk, Darrell D. Ebbing

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