

## FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

| Elements              | metal ion | non-metal ion | Chemical Formula | Name              |
|-----------------------|-----------|---------------|------------------|-------------------|
| potassium and iodine  | $K^{+1}$  | $I^{-1}$      | KI               | Potassium Iodide  |
| barium and chlorine   | $Ba^{2+}$ | $Cl^{-1}$     | $BaCl_2$         | Barium Chloride   |
| sodium and oxygen     | $Na^{+1}$ | $O^{2-}$      | $Na_2O$          | sodium Oxide      |
| aluminum and nitrogen | $Al^{3+}$ | $N^{3-}$      | AlN              | Aluminum Nitride  |
| lithium and sulfur    | $Li^{+1}$ | $S^{2-}$      | $Li_2S$          | Lithium Sulfide   |
| magnesium and oxygen  | $Mg^{+2}$ | $O^{2-}$      | MgO              | Magnesium Oxide   |
| calcium and chlorine  | $Ca^{2+}$ | $Cl^{-1}$     | $CaCl_2$         | calcium chloride  |
| boron and chlorine    | $B^{3+}$  | $Cl^{-1}$     | $BCl_3$          | boron chloride    |
| beryllium and sulfur  | $Be^{2+}$ | $S^{2-}$      | BeS              | beryllium sulfide |

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

|           | F       | Cl       | $O^{2-}$          | $S^{2-}$  | $N^{3-}$  | $P^{3-}$  |
|-----------|---------|----------|-------------------|-----------|-----------|-----------|
| $Li^{+}$  | LiF     | LiCl     | $Li_2O$           | $Li_2S$   | $Li_3N$   | $Li_3P$   |
| $Na^{+}$  | NaF     | NaCl     | Na <sub>2</sub> O | $Na_2S$   | $Na_3N$   | $Na_3P$   |
| $K^{+}$   | KF      | KCl      | $K_2O$            | $K_2S$    | $K_3N$    | $K_3P$    |
| $Be^{2+}$ | $BeF_2$ | $BeCl_2$ | BeO               | BeS       | $Be_3S_2$ | $Be_3P_2$ |
| $Mg^{2+}$ | $MgF_2$ | $MgCl_2$ | MgO               | MgS       | $Mg_3N_2$ | $Mg_3P_2$ |
| $Ca^{2+}$ | $CaF_2$ | $CaCl_2$ | CaO               | CaS       | $Ca_3N_2$ | $Ca_3P_2$ |
| $Al^{3+}$ | $AlF_3$ | $AlCl_3$ | $Al_2O_3$         | $Al_2S_3$ | AlN       | AlP       |
| $Ga^{3+}$ | $GaF_3$ | $GaCl_3$ | $Ga_2O_3$         | $Ga_2S_3$ | GaN       | GaP       |
| $B^{3+}$  | $BF_3$  | $BCl_3$  | $B_2O_3$          | $B_2S_3$  | BN        | BP        |

# Ionic Formulas Questions Chemthink

**Jin-Ying Zhang**



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