

# CHARGES, NAMES, AND FORMULAS OF COMMON IONS

## Positive Ions (Cations)

### +1 Charge

Name	Formula
Ammonium	$\text{NH}_4^+$
Copper(I)	$\text{Cu}^+$
Lithium	$\text{Li}^+$
Potassium	$\text{K}^+$
Silver	$\text{Ag}^+$
Sodium	$\text{Na}^+$

### +2 Charge

Name	Formula
Barium	$\text{Ba}^{2+}$
Calcium	$\text{Ca}^{2+}$
Copper(II)	$\text{Cu}^{2+}$
Iron(II)	$\text{Fe}^{2+}$
Lead(II)	$\text{Pb}^{2+}$
Magnesium	$\text{Mg}^{2+}$
Nickel(II)	$\text{Ni}^{2+}$
Strontium	$\text{Sr}^{2+}$
Tin(II)	$\text{Sn}^{2+}$
Zinc	$\text{Zn}^{2+}$

### +3 Charge

Name	Formula
Aluminum	$\text{Al}^{3+}$
Iron(III)	$\text{Fe}^{3+}$

### +4 Charge

Name	Formula
Lead(IV)	$\text{Pb}^{4+}$
Tin(IV)	$\text{Sn}^{4+}$

## Negative Ions (Anions)

### -1 Charge

Name	Formula
Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
Bromide	$\text{Br}^-$
Chloride	$\text{Cl}^-$
Dihydrogen Phosphate	$\text{H}_2\text{PO}_4^-$
Fluoride	$\text{F}^-$
Hydroxide	$\text{OH}^-$
Hydrogen Carbonate (bicarbonate)	$\text{HCO}_3^-$
Hydrogen Sulfate (bisulfate)	$\text{HSO}_4^-$
Hypochlorite	$\text{ClO}^-$
Iodide	$\text{I}^-$
Nitrate	$\text{NO}_3^-$
Permanganate	$\text{MnO}_4^-$
Thiocyanate	$\text{SCN}^-$

### -2 Charge

Name	Formula
Carbonate	$\text{CO}_3^{2-}$
Chromate	$\text{CrO}_4^{2-}$
Dichromate	$\text{Cr}_2\text{O}_7^{2-}$
Hydrogen Phosphate	$\text{HPO}_4^{2-}$
Oxide	$\text{O}^{2-}$
Oxalate	$\text{C}_2\text{O}_4^{2-}$
Sulfate	$\text{SO}_4^{2-}$
Sulfite	$\text{SO}_3^{2-}$

### -3 Charge

Name	Formula
Nitride	$\text{N}^{3-}$
Phosphate	$\text{PO}_4^{3-}$

# Ions And Their Chemical Formula Answers

**William Crowell Bray, Wendell Mitchell  
Latimer**



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