

Ionic & Covalent Compound Naming**Race**

First, identify whether these compounds are ionic or covalent. Then, use the correct formula writing rules to write the correct chemical formulas for each compound.

1)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	copper (II) chlorite		Cu(ClO)
2)	sodium hydroxide		NaOH
3)	nitrogen dioxide		NO ₂
4)	cobalt (III) oxalate		Co ₂ (C ₂ O ₄) ₃
5)	ammonium sulfide		(NH ₄) ₂ S
6)	aluminum cyanide		Al(CN) ₃
7)	carbon disulfide		CS ₂
8)	tetraphosphorous pentoxide		P ₄ O ₅
9)	potassium permanganate		KMnO ₄
10)	manganese (III) chloride		MnCl ₃
11)	Compound Name	Type of Compound: Ionic or Covalent	Chemical Formula
	calcium bromate		Ca(BrO ₃) ₂
12)	carbon monoxide		CO
13)	potassium oxide		K ₂ O
14)	antimony tribromide		SbBr ₃
15)	zinc phosphate		Zn ₃ (PO ₄) ₂
16)	copper (II) bicarbonate		Cu(HCO ₃) ₂
17)	dinitrogen tetroxide		N ₂ O ₄

Ionic And Covalent Nomenclature Ws Answers

Daniela Niemeyer



Ionic And Covalent Nomenclature Ws Answers:

Reviewing **Ionic And Covalent Nomenclature Ws Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Ionic And Covalent Nomenclature Ws Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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