

### Honors Chemistry Review Problems

1. What is the wavelength of X rays having a frequency of  $4.80 \times 10^{17}$  Hz?
2. A helium-neon laser emits light with a wavelength of 633 nm. What is the frequency of this light?
3. An FM radio station broadcasts at a frequency of 98.5 MHz. What is the wavelength of the station's broadcast signal?
4. Calculate the energy of a gamma ray photon whose frequency is  $5.02 \times 10^{20}$  Hz.
5. What is the difference in energy between a photon of violet light with a frequency of  $6.8 \times 10^{14}$  Hz and a photon of red light with a frequency of  $4.3 \times 10^{14}$  Hz?
7. Calculate the energy of a photon of ultraviolet light that has a wavelength of 49.0 nm.
8. Orange light has a frequency of  $4.8 \times 10^{14}$  Hz. What is the energy of one quantum of orange light?
9. New-generation cordless phones use a  $9.00 \times 10^2$  MHz frequency and can be operated up to 60.0 m from their base. How many wavelengths of the electromagnetic waves can fit between your ear and a base 60.0 m away?
10. The highest *directly* measured frequency is  $5.20 \times 10^{14}$  Hz, corresponding to one of the transitions in iodine-127. How many wavelengths of electromagnetic waves with this frequency could fit across a dot on a book.  
(Assume the dot is  $2.00 \times 10^{-4}$  m in diameter.)

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**Arturo Cuomo**



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