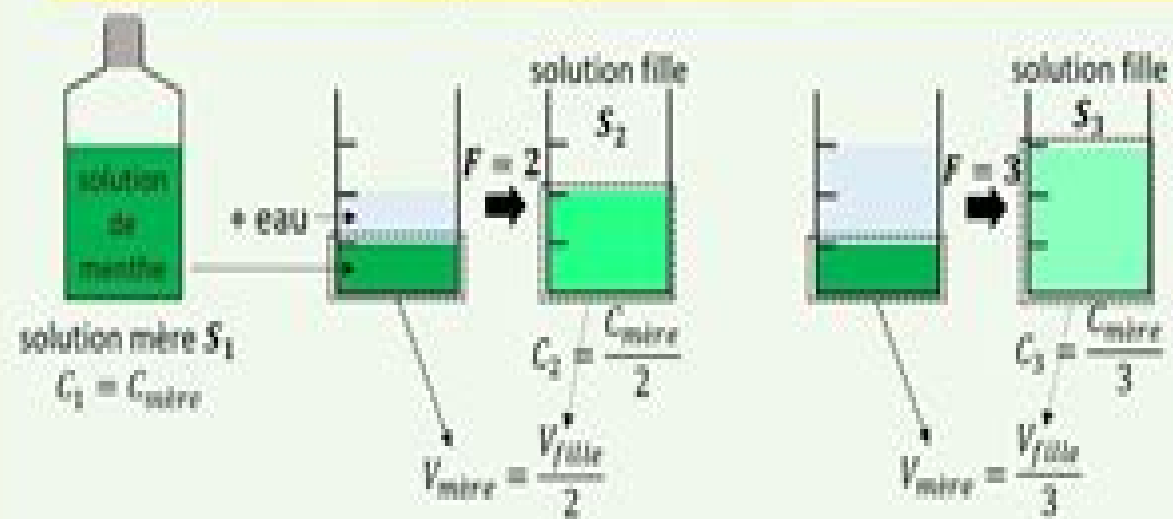


La dilution pour préparer une solution

diluer = rajouter du solvant (de l'eau) à une solution \Rightarrow diminuer la concentration en soluté

rappel de math !

$\frac{a}{b} \times \frac{c}{d}$



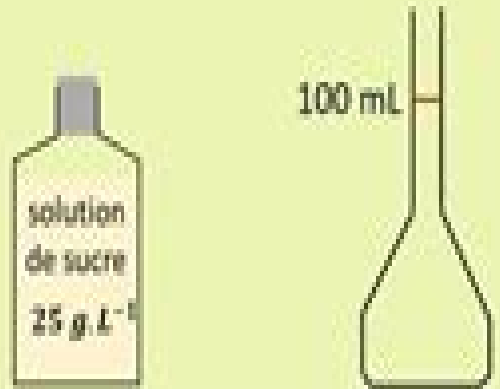
$C_{fille} = \frac{C_{mère}}{F}$
 $V_{mère} = \frac{V_{fille}}{F}$

factor de dilution
nombre de fois qu'on dilue

Application : À disposition : une solution mère de sucre à 25 g.L^{-1} . On souhaite préparer 100 mL de solution fille à 5 g.L^{-1} .

- solution fille à préparer dans une fiole jaugée de 100 mL $\Rightarrow V_{fille} = 100 \text{ mL}$
- $V_{mère}$ à introduire dans la fiole ?

$\rightarrow V_{mère} = \frac{V_{fille}}{F}$



Le Principe De Sinceacuteriteacute Budgeacutetaire

Michael Hoelscher



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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction

with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accommodate their electrons demonstration of creating the p arenium cation by nitration of bromobenzene

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