

KISA Pre Board Examination - Answer key

SUBJECT-CHEMISTRY

Section A

Question 1.

1x15=15M

- i. (a). they have similar electronic configuration in valence shell.
- ii. (c). CuSO_4
- iii. (b). Oxygen is released at anode.
- iv. (c). Substitution
- v. (a). He
- vi. (c). Consists of molecules
- vii. (b). Lead
- viii. (d). Argon
- ix. (a). Magnesium oxide
- x. (b). Aqueous acetic acid
- xi. (c). Reacts to form water, nitrogen dioxide and carbon dioxide
- xii. (a). K_2SO_4
- xiii. (c). 1
- xiv. (c). Fe, Cr and Ni
- xv. (a). Decreases

Question 2.

- (i) (a) A-----→ Conc. sulphuric acid (3)
B-----→ sodium nitrate/potassium nitrate
C-----→ Nitric acid



- (ii) Column A Column B 1x5=5M

- a) Sodium chloride — ionic bond
- b) Ammonium ion — covalent and co-ordinate bond
- c) Electronegativity across the period —I increases
- d) Nonmetallic character down the group — decreases

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Table of Contents Jeemainpaper2 Ans Key 2014 Code K

1. Understanding the eBook Jeemainpaper2 Ans Key 2014 Code K
 - The Rise of Digital Reading Jeemainpaper2 Ans Key 2014 Code K
 - Advantages of eBooks Over Traditional Books
2. Identifying Jeemainpaper2 Ans Key 2014 Code K
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jeemainpaper2 Ans Key 2014 Code K
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jeemainpaper2 Ans Key 2014 Code K
 - Personalized Recommendations
 - Jeemainpaper2 Ans Key 2014 Code K User Reviews and Ratings
 - Jeemainpaper2 Ans Key 2014 Code K and Bestseller Lists
5. Accessing Jeemainpaper2 Ans Key 2014 Code K Free and Paid eBooks
 - Jeemainpaper2 Ans Key 2014 Code K Public Domain eBooks
 - Jeemainpaper2 Ans Key 2014 Code K eBook Subscription Services
 - Jeemainpaper2 Ans Key 2014 Code K Budget-Friendly Options
6. Navigating Jeemainpaper2 Ans Key 2014 Code K eBook Formats
 - ePub, PDF, MOBI, and More
 - Jeemainpaper2 Ans Key 2014 Code K Compatibility with Devices
 - Jeemainpaper2 Ans Key 2014 Code K Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jeemainpaper2 Ans Key 2014 Code K
 - Highlighting and Note-Taking Jeemainpaper2 Ans Key 2014 Code K
 - Interactive Elements Jeemainpaper2 Ans Key 2014 Code K
8. Staying Engaged with Jeemainpaper2 Ans Key 2014 Code K

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Jeemainpaper2 Ans Key 2014 Code K
- 9. Balancing eBooks and Physical Books Jeemainpaper2 Ans Key 2014 Code K
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jeemainpaper2 Ans Key 2014 Code K
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jeemainpaper2 Ans Key 2014 Code K
 - Setting Reading Goals Jeemainpaper2 Ans Key 2014 Code K
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jeemainpaper2 Ans Key 2014 Code K
 - Fact-Checking eBook Content of Jeemainpaper2 Ans Key 2014 Code K
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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