

## Grade 10 Life Sciences

### Class test 1

Date:

#### Section A

##### Question 1

Give the correct **biological term** for each of the following descriptions. Write only the term next to the question number (1.1 to 1.5) in the ANSWER BOOK.

- 1.1 The variety of living organisms on earth.
- 1.2 A group of food chains arranged to show how the feeding of organisms is connected.
- 1.3 Factors that relate to the physical features of the land.
- 1.4 An area with a particular climate and soil type that influences the plants and animals that live there.
- 1.5 Organisms that feed on dead or decaying matter. (5 x 1) (5)

**TOTAL SECTION A: [5]**

#### Section B

##### Question 2

- 2.1 The diagram below represents a food chain in a garden.



- 2.1.1 Name the organism above that represents a:

(a) Carnivore

(1)

# Life Science Question Paper Grade 1term 3 201

**Sebastian Brünink**



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*Cornell University Description of Courses* Cornell University, 1976      **Resources in Education** ,1982      Science Books & Films ,1980      **Primary Education, Popular Educator** ,1926      **The Athenæum** James Silk Buckingham, John Sterling, Frederick Denison Maurice, Henry Stebbing, Charles Wentworth Dilke, Thomas Kibble Hervey, William Hepworth Dixon, Norman Maccoll, Vernon Horace Rendall, John Middleton Murry, 1882      **The Athenæum** ,1878      *The Athenæum* ,1893      **Scientific American** ,1903      **Harcourt Science: Physical science, [grade] 4, Units E and F, teacher's ed. [v. 18]. Life science, [grade] 5, Units A and B, teacher's ed** ,2000      **English Mechanic and Mirror of Science** ,1886  
English Mechanic and Mirror of Science and Art ,1886      New England Journal of Education Thomas Williams Bicknell, Albert Edward Winship, Anson Wood Belding, 1877      *Journal of Education* ,1881      **English Mechanics and the World of Science** ,1889      **English Mechanic and World of Science** ,1888      **Undergraduate and Graduate Courses and Programs** Iowa State University, 2009      **American Miller** ,1912      *The Americana* ,1923      *Normal Instructor and Teachers World* ,1902      **Journal of the Engineering Institute of Canada** ,1920 Vol 7 no 7 July 1924 contains papers prepared by Canadian engineers for the first World power conference July 1924

## Whispering the Techniques of Language: An Emotional Journey through **Life Science Question Paper Grade 1term 3 201**

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is  $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13  $2\text{Ag}^+_{\text{aq}} + 2\text{F}^-_{\text{aq}} + 2\text{NH}_4^+_{\text{aq}} + \text{Cr}_2\text{O}_7^{2-}_{\text{aq}} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7\text{s} + 2\text{NH}_4^+_{\text{aq}} + 2\text{F}^-_{\text{aq}}$  because two  $\text{NH}_4^+_{\text{aq}}$

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