Borazine, also known as inorganic benzene, can be prepared by the reaction of 3-equivalents of "X" with 6-equivalents of "Y". "X" and "Y", respectively are :

 \bigcirc B(OH)₃ and NH₃

 $egin{array}{c} egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}{c} \egin{array}$

 \bigcirc $\mathbf{B_2H_6}$ and $\mathbf{HN_3}$

 \bigcirc NH $_3$ and B $_2$ O $_3$

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