

1 Generally transition elements and their salts are coloured due to the presence of unpaired electrons in metal ions. Which of the following compounds are coloured?

- KMnO₄
- Ce(SO₄)₂
- TiCl₃
- Cu₂Cl₂

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

2 General electronic configuration of actinoids is (n-2)f¹⁻¹⁴ (n-1)d⁰⁻² ns². Which of the following actinoids have one electron in 6d orbital?

- U (Atomic no. 92)
- Np (Atomic no. 93)
- Pu (Atomic no. 94)
- Am (Atomic no. 95)

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

3 Which of the following lanthanoids show +2 oxidation state besides the characteristic oxidation state +3 of lanthanoids?

- Ce
- Eu
- Yb
- Ho

Choose the correct option

- (a, b)
- (b, c)
- (c, d)
- (a, d)

4 On heating lead nitrate forms oxides of nitrogen and lead. The oxides formed are-

- N₂O, PbO
- NO₂, PbO
- NO, PbO
- NO, PbO₂

5 A brown ring is formed in the ring test for NO₃⁻ ion. It is due to the formation of-

- [Fe (H₂O)₅ (NO)]²⁺
- FeSO₄, NO₂
- [Fe (H₂O)₄ (NO)₂]²⁺
- FeSO₄, HNO₃

6 The oxidation state of the central atom in the anion of the compound NaH₂ PO₂ will be-

- +3
- +5
- +1
- 3

7 A black compound of manganese reacts with a halogen acid to give greenish-yellow gas. When an excess of this gas reacts with NH₃ an unstable trihalide is formed. In this process, the oxidation state of nitrogen changes from-

- 3 to +3
- 3 to 0
- 3 to +5
- 0 to -3

8 Which of the following is isoelectronic pair?

- ICl₂, ClO₂
- BrO₂⁻, BrF₂⁺
- ClO₂, BrF
- CN⁻, O₃

9 Which of the following statements are correct for SO₂ gas?

- It acts as a bleaching agent in moist conditions.
 - Its molecule has linear geometry.
 - Its dilute solution is used as a disinfectant.
 - It can be prepared by the reaction of dilute H₂SO₄ with metal sulphide.
- (a, b)
 - (b, c)
 - (c, d)
 - (a, c)

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