

# CHEMISTRY

## QUESTION PAPER

## ANSWERS

## CLASS 12 HSC

## MAHARASTRA BOARD

20. Given data :
- (a)  $R = 8.314 \text{ J/K mol}$
  - (b) Atomic mass  $X = 12$
  - (c)  $K_f$  for water =  $1.86 \text{ K kg mol}^{-1}$
  - (d)  $1^\circ = 10000^\circ$
  - (e)  $N_A = 6.022 \times 10^{23}$

### SECTION - A

Q. 1. Select and write the correct answer for the following multiple choice type of questions :

- (a) Solubility of silver is NOT observed in \_\_\_\_\_  
 (a)  $\text{NaCl}$  (b)  $\text{KCl}$   
 (c)  $\text{AgCl}$  (d)  $\text{NaCl}$
- (b) The freezing point of 1 in aqueous solution of urea, if  $K_f$  for water is  $1.86 \text{ K kg mol}^{-1}$  is \_\_\_\_\_  
 (a)  $1.86^\circ\text{C}$  (b)  $-1.86^\circ\text{C}$   
 (c)  $0.186^\circ\text{C}$  (d)  $-0.186^\circ\text{C}$
- (c) Green layer is displayed by \_\_\_\_\_  
 (a)  $\text{NO}$  (b)  $\text{NO}_2$   
 (c)  $\text{NO}_3$  (d)  $\text{N}_2\text{O}_5$
- (d) When excess of  $\text{AgNO}_3$  is added to a complex, one mole of  $\text{AgCl}$  is precipitated. The formula of complex is \_\_\_\_\_  
 (a)  $[\text{Co}(\text{Cl})_2(\text{OH})_2] \cdot 2\text{H}_2\text{O}$  (b)  $[\text{CoCl}(\text{OH})_2] \cdot 2\text{H}_2\text{O}$   
 (c)  $[\text{Co}(\text{Cl})_2(\text{OH})_2] \cdot 2\text{H}_2\text{O}$  (d)  $[\text{Co}(\text{OH})_2] \cdot 2\text{H}_2\text{O}$
- (e) The value of  $n_e$  for the oxidation of 1 mole of sulphur dioxide to sulphur trioxide is \_\_\_\_\_  
 (a) 2 (b) 1  
 (c) 4 (d) 3

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