## Analytical Chemistry MCQ Test. September'21.

- The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
- ammonium hydroxide.

  1. Cu(OH): 2 Zn(OH): 3 Fe(OH): 4 None
- 2. The salt solution which does not react with ammonium hydroxide is ------
- 1. Calcium nitrate 2. Zinc nitrate
- 3.Lead nitrate 4. Copper nitrate
- 3. The metal oxide which can react with acids as well as alkali is------
  - 1. Silver oxide 2. Copper (II) oxide
  - 3. Aluminum oxide 4. Calcium oxide
  - 4.Al metal shows amphoteric behaviour.it can be shown by
  - 1. Its reaction with dil.NaOH only.
  - Its reaction with dil.HCl only.
  - 3. Its reaction with both dil.HCl and dil.NaOH.
  - It does not react with either dil.HCl or dil.NaOH

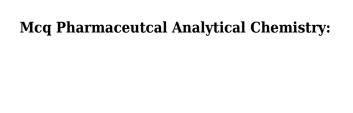
#### QUESTION 5

Which of the following hydroxides soluble in excess of NaOH?

- 1. Cu (OH)<sub>2</sub> 2. Zn (OH)<sub>2</sub> 3. Fe (OH)<sub>2</sub> 4Fe(OH)<sub>3</sub>
- The salt which gives a dirty green precipitate with sodium hydroxide solution is.
  - 1. Iron (III) sulphate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
  - 1. Iron (III) sulphate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
  - 1. Iron (III) choride 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
  - 1. Lead nitrate 2. Iron (II) sulphate
  - 3.Copper sulphate 4. Zinc sulphate

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**Jin-Ying Zhang** 



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#### **Table of Contents Mcq Pharmaceutcal Analytical Chemistry**

- 1. Understanding the eBook Mcq Pharmaceutcal Analytical Chemistry
  - The Rise of Digital Reading Mcq Pharmaceutcal Analytical Chemistry
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Mcq Pharmaceutcal Analytical Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcq Pharmaceutcal Analytical Chemistry
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mcq Pharmaceutcal Analytical Chemistry
  - Personalized Recommendations
  - Mcq Pharmaceutcal Analytical Chemistry User Reviews and Ratings
  - Mcg Pharmaceutcal Analytical Chemistry and Bestseller Lists
- 5. Accessing Mcg Pharmaceutcal Analytical Chemistry Free and Paid eBooks
  - Mcg Pharmaceutcal Analytical Chemistry Public Domain eBooks
  - Mcq Pharmaceutcal Analytical Chemistry eBook Subscription Services
  - Mcq Pharmaceutcal Analytical Chemistry Budget-Friendly Options
- 6. Navigating Mcg Pharmaceutcal Analytical Chemistry eBook Formats

- o ePub, PDF, MOBI, and More
- Mcq Pharmaceutcal Analytical Chemistry Compatibility with Devices
- Mcq Pharmaceutcal Analytical Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mcg Pharmaceutcal Analytical Chemistry
  - Highlighting and Note-Taking Mcq Pharmaceutcal Analytical Chemistry
  - Interactive Elements Mcq Pharmaceutcal Analytical Chemistry
- 8. Staying Engaged with Mcq Pharmaceutcal Analytical Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcq Pharmaceutcal Analytical Chemistry
- 9. Balancing eBooks and Physical Books Mcq Pharmaceutcal Analytical Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcq Pharmaceutcal Analytical Chemistry
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mcg Pharmaceutcal Analytical Chemistry
  - Setting Reading Goals Mcq Pharmaceutcal Analytical Chemistry
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcq Pharmaceutcal Analytical Chemistry
  - Fact-Checking eBook Content of Mcq Pharmaceutcal Analytical Chemistry
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
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