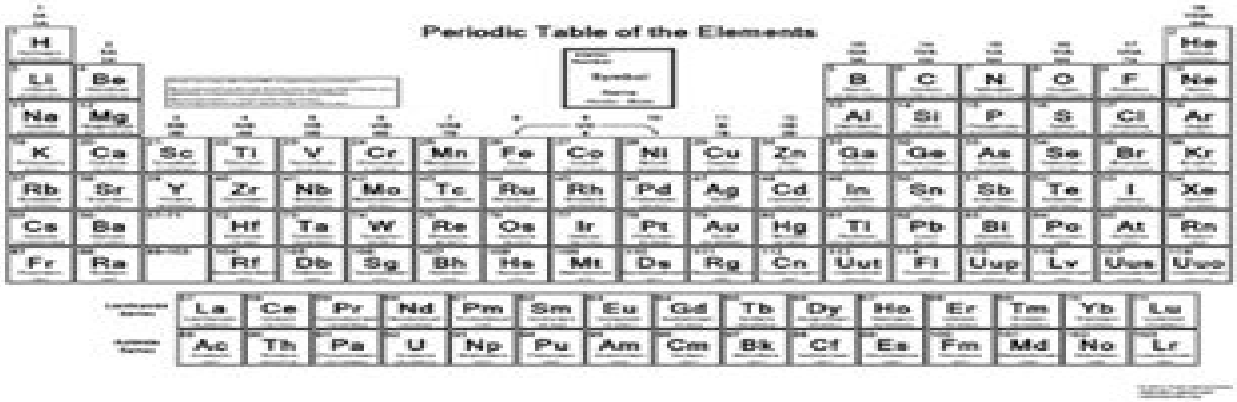
Chemistry Study Guide



🔬 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🧑‍🔬✨.

💡 Matter:

Anything that has **mass** and **occupies space** 🏋️🌍

Examples: air, gases, liquids, solidz!

⚛️ Atoms:

Atoms are the **smallest, indivisible particle** 🔬✂️.

- **Outer part** → *shell/orbit/energy level* → carries the electronz (e−)
- **Inner part** → *nucleus* → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the **subatomic particles**

🧪 Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~0	-1	Orbit/Shell

Mcquarrie General Chemistry Study Guide

R Barnett

Mcquarrie General Chemistry Study Guide:

Whispering the Techniques of Language: An Psychological Journey through **Mcquarrie General Chemistry Study Guide**

In a digitally-driven earth where displays reign great and immediate interaction drowns out the subtleties of language, the profound strategies and psychological subtleties concealed within phrases frequently go unheard. Yet, nestled within the pages of **Mcquarrie General Chemistry Study Guide** a interesting fictional prize blinking with organic thoughts, lies an exceptional quest waiting to be undertaken. Published by an experienced wordsmith, this marvelous opus invites readers on an introspective journey, softly unraveling the veiled truths and profound influence resonating within the very cloth of each word. Within the psychological depths of this poignant evaluation, we shall embark upon a honest exploration of the book is core subjects, dissect its fascinating publishing style, and fail to the powerful resonance it evokes strong within the recesses of readers hearts.

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