

1. As a solute is added to a solvent, what happens to the freezing point and the boiling point of the solution?
- ☒ A) The freezing point decreases and the boiling point increases.
- B) The freezing point increases and the boiling point increases.
- C) The freezing point increases and the boiling point decreases.
- D) The freezing point decreases and the boiling point decreases.
2. Compared to the freezing point of 1.0 M KCl(aq) at standard pressure, the freezing point of 1.0 M CaCl<sub>2</sub>(aq) at standard pressure is
- More ions*  
☒ A) lower  
 B) higher  
 C) the same
3. Which aqueous solution of KI freezes at the lowest temperature?
- A) 1 mol of KI in 500. g of water
- ☒ B) 2 mol of KI in 500. g of water *(more concentrated)*
- C) 1 mol of KI in 1000. g of water
- D) 2 mol of KI in 1000. g of water
4. A 1 kilogram sample of water will have the highest freezing point when it contains
- A)  $1 \times 10^{21}$  dissolved particles
- ☒ B)  $1 \times 10^{17}$  dissolved particles
- C)  $1 \times 10^{19}$  dissolved particles
- D)  $1 \times 10^{23}$  dissolved particles *lowest f.p.*
5. A solution consists of 0.50 mole of CaCl<sub>2</sub> dissolved in 100. grams of H<sub>2</sub>O at 25°C. Compared to the boiling point and freezing point of 100. grams of H<sub>2</sub>O at standard pressure, the solution at standard pressure has
- ☒ A) a higher boiling point and a lower freezing point
- B) a lower boiling point and a lower freezing point
- C) a higher boiling point and a higher freezing point
- D) a lower boiling point and a higher freezing point
6. Which solution will freeze at the lowest temperature?
- A) 1 mole of sugar in 500 g of water
- B) 1 mole of sugar in 1,000 g of water
- C) 2 moles of sugar in 1,000 g of water
- ☒ D) 2 moles of sugar in 500 g of water
7. Which solution has the highest boiling point at standard pressure?
- 4 ions*  
☒ A) 0.10 M K<sub>3</sub>PO<sub>4</sub>(aq) *3 K<sup>+</sup> 3 PO<sub>4</sub><sup>-</sup>*
- B) 0.10 M KCl(aq)
- C) 0.10 M KNO<sub>3</sub>(aq)
- D) 0.10 M K<sub>2</sub>SO<sub>4</sub>(aq)
8. Which concentration of a solution of CH<sub>3</sub>OH in water has the lowest freezing point?
- A) 0.001 M
- ☒ B) 0.1 M
- C) 0.01 M
- D) 0.0001 M
9. How do the boiling point and freezing point of a solution of water and calcium chloride at standard pressure compare to the boiling point and freezing point of water at standard pressure?
- A) Both the freezing point and boiling point of the solution are higher.
- B) The freezing point of the solution is higher and the boiling point of the solution is lower.
- ☒ C) The freezing point of the solution is lower and the boiling point of the solution is higher.
- D) Both the freezing point and boiling point of the solution are lower.
10. Compared to a 2.0 M aqueous solution of NaCl at 1 atmosphere, a 3.0 M aqueous solution of NaCl at 1 atmosphere has a
- A) lower boiling point and a higher freezing point
- B) higher boiling point and a higher freezing point
- C) lower boiling point and a lower freezing point
- ☒ D) higher boiling point and a lower freezing point



# June 2005 Chemistry Regents Answers

**RJ Shavelson**





## June 2005 Chemistry Regents Answers:

*Chemistry and Industry*, 2009      Testing Regimes, Accountabilities and Education Policy Bob Lingard, Goli Rezai-Rashti, Wayne Martino, 2017-10-02 Around the globe various kinds of testing including high stakes national census testing have become meta policies steering educational systems in particular directions and having great effects on schools and on teacher practices as well as upon student learning and curricula There has also been a complementary global aspect to this with the OECD's PISA and IEA's TIMSS and PIRLS which have had impacts on national education systems and their policy frameworks While there has been a globalized educational policy discourse that suggests that high stakes standardised testing will drive up standards and enhance the quality of a nation's human capital and thus their international economic competitiveness this discourse still manifests itself in specific vernacular path dependent ways in different nations High stakes testing and its effects can also be seen as part of the phenomenon of the datafication of the world and policy as numbers linked to other reforms of the state including new public management network governance and top down and test based modes of accountability This edited collection provides theoretically and empirically informed analyses of these developments This book was originally published as a special issue of the *Journal of Education Policy*      Research and Development, 2004      *El-Hi Textbooks & Serials in Print*, 2005, 2005      **R & D**, 2005      **New York Laws Governing Business Entities, Annotated** New York (State), 2008      Research & Development, 2005-07      **Who's Who in Science and Engineering 2008-2009** Who's Who Marquis, Marquis Who's Who, 2007-12      *Supplementary Education* Edmund W. Gordon, Beatrice L. Bridgall, Aundra Saa Meroe, 2005 In *Supplementary Education* the editors argue that while access to schools that enable and expect academic achievement is a necessary ingredient for the education of students schools alone may not be sufficient to ensure universally high levels of academic development Supplemental educational experiences may also be needed The idea of supplementary education is based on the assumption that high academic achievement is closely associated with exposure to family and community based activities and learning experiences that occur both in and out of school in support of academic learning For low income and some ethnic minority student groups opportunities to participate in such activities are generally under resourced and underutilized in comparison to the access to and participation in such activities by many European and Asian Americans from mid to high socio economic backgrounds This book makes the case for supplementary education Specifically it focuses on the need for universal access to high levels of academic achievement and the challenge of reducing the achievement gap that exists between Asian American and European American students and their African American Latina o and Native American counterparts Having posed the problem the editors define the construct and provide in depth descriptions of some of the more colloquial expressions of supplementation in after school care youth development and other forms of supplemental education The editors close with a discussion of the emerging institutionalization and need for more thoughtful and rigorous research of the supplementary education movement      **The**



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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**June 2005 Chemistry Regents Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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