

$$LE = -635 - (178 + 249 + 590 + 1145 + (-141) + 798) = -3454$$

$$\begin{array}{r} -3736 \\ +3454 \\ \hline \end{array}$$

-1858

42184

-2184

answer = kJ/mol

e) Describe and explain the factors that affect the values of lattice enthalpies.

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