

**JIPMER-02-06-2019-Morning**  
**Chemistry**

1.  $^{19}\text{F}^{-1}$ ,  $^{16}\text{O}^{-2}$ ,  $^{20}\text{Ne}$  choose the correct statement
  - (1) Both  $\text{O}^{-2}$  and  $\text{F}^{-}$  are isoelectronic.
  - (2) All given have equal no of  $e^{-}$
  - (3)  $\text{F}^{-}$  and  $\text{Ne}$  have equal number of  $e^{-}$
  - (4) All
  
2. Hund's Rule state that:
  - (1) Number of two  $e^{-}$  can be in two separate orbitals
  - (2) Number of two  $e^{-}$  can be present with similar spin in a orbital.
  - (3) No one  $e^{-}$  can exist in 'd' orbital
  - (4) None of these
  
3. If two atoms have equal number of electron it is called:
  - (1) Iso-electronic
  - (2) isotone
  - (3) isobar
  - (4) None of these

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## **Table of Contents Jipmer 2mbbs Entrance Question Paper**

1. Understanding the eBook Jipmer 2mbbs Entrance Question Paper
  - The Rise of Digital Reading Jipmer 2mbbs Entrance Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Jipmer 2mbbs Entrance Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Jipmer 2mbbs Entrance Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Jipmer 2mbbs Entrance Question Paper
  - Personalized Recommendations
  - Jipmer 2mbbs Entrance Question Paper User Reviews and Ratings
  - Jipmer 2mbbs Entrance Question Paper and Bestseller Lists
5. Accessing Jipmer 2mbbs Entrance Question Paper Free and Paid eBooks
  - Jipmer 2mbbs Entrance Question Paper Public Domain eBooks
  - Jipmer 2mbbs Entrance Question Paper eBook Subscription Services
  - Jipmer 2mbbs Entrance Question Paper Budget-Friendly Options

6. Navigating Jipmer 2mbbs Entrance Question Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Jipmer 2mbbs Entrance Question Paper Compatibility with Devices
  - Jipmer 2mbbs Entrance Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Jipmer 2mbbs Entrance Question Paper
  - Highlighting and Note-Taking Jipmer 2mbbs Entrance Question Paper
  - Interactive Elements Jipmer 2mbbs Entrance Question Paper
8. Staying Engaged with Jipmer 2mbbs Entrance Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Jipmer 2mbbs Entrance Question Paper
9. Balancing eBooks and Physical Books Jipmer 2mbbs Entrance Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Jipmer 2mbbs Entrance Question Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Jipmer 2mbbs Entrance Question Paper
  - Setting Reading Goals Jipmer 2mbbs Entrance Question Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Jipmer 2mbbs Entrance Question Paper
  - Fact-Checking eBook Content of Jipmer 2mbbs Entrance Question Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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