

Mole and Stoichiometry Mini-Lab

Calculations: Show **ALL** work (bridge/grid).

The **GIVEN** for calculations #4 and #5 should be measurements, not results from previous calculations.

4. How many moles of water are in the graduated cylinder?
UNKNOWN: mol H_2O GIVEN: g H_2O (data table)
5. How many molecules of water are in the graduated cylinder?
UNKNOWN: molecules H_2O
GIVEN: g H_2O (data table)
6. For an experiment, 4.75 moles of water are needed.
 - a) How many grams of water are needed?
 - b) How many milliliters of water would this be?
UNKNOWN: g H_2O GIVEN: 4.75 mol H_2O

UNKNOWN: mL H_2O GIVEN: 4.75 mol H_2O

Mini Lab 11 Apply Stoichiometry

**James Silk Buckingham, John
Sterling, Frederick Denison
Maurice, Henry Stebbing, Charles
Wentworth Dilke, Thomas Kibble
Hervey, William Hepworth
Dixon, Norman Maccoll, Vernon Horace
Rendall, John Middleton Murry**

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Improving Student Comprehension of Stoichiometric Concepts Connie Lynn Bannick Kemner, 2007 **Nuclear Science Abstracts** ,1961 *TID.* ,19?? *General Chemistry Laboratory Text with Qualitative Analysis* Wilbert Hutton, 1968
Energy Research Abstracts ,1982 **Addison-Wesley Small-scale Chemistry** Dennis D. Staley, Edward L. Waterman, 1995 **Chemical News and Journal of Industrial Science** ,1920 *The Athenaeum* ,1905 *Laboratory Apparatus and Reagents Selected for Laboratories of Chemistry and Biology* Thomas, Arthur H., Company, Philadelphia, 1914
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