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**Rachel Sandford**



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co}^{2+} \rightarrow \text{Fe}^{2+} + \text{Co}$   
oxidation  $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$  reduction  $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$  24 3  $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ag}^{+} + \text{Ni}$  3  $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ag}^{+} + \text{Ni}$  25  $\text{Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$  oxidation reduction 26  $\text{O}_2$

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