

## 13.1

## THE NATURE OF GASES

## Section Review

## Objectives

1. describe the characteristics of matter having gaseous properties
2. distinguish between a gas and vapour
3. define the macroscopic and kinetic approaches to understanding gases

## Vocabulary

- |                 |                         |                  |
|-----------------|-------------------------|------------------|
| a. condensing   | b. vacuum               | c. pressure      |
| d. condensation | e. atmospheric pressure | f. absolute zero |
| g. properties   | h. diffusion            |                  |

## Part A Completion

Use the vocabulary terms in parentheses to complete each of the sentences. Write the complete sentence on the lines below.

Diffusion shows that the \_\_\_\_\_ of a gas is moving continuously in random motion. The \_\_\_\_\_ of the gas is constant for a given volume and temperature. \_\_\_\_\_ the gas particles are moving. At \_\_\_\_\_, all molecular motion ceases to exist. \_\_\_\_\_ matter has \_\_\_\_\_ characteristics. In particular, there is neither a definite \_\_\_\_\_ nor a definite \_\_\_\_\_ of particles. Gases expand to fill the container in which they are placed. Gases exert \_\_\_\_\_ on the walls of the container. Gases are \_\_\_\_\_ and \_\_\_\_\_ for \_\_\_\_\_.

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## Part B True-False

Write **T** for true and **F** for false. Write **OT** for other than **OT** or **OT** for other than **OT**.

- \_\_\_\_\_ 11. Diffusion is a process that occurs in gases.
- \_\_\_\_\_ 12. Gases are compressible.

# Nature Of Gases Section Review Answer Key

**Clifford Lane**



## **Nature Of Gases Section Review Answer Key:**

## Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Nature Of Gases Section Review Answer Key**

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