

Unit

12

Hydrocarbons

Long Answer Questions

Q.1 What are hydrocarbons? Give their different classes depending upon their structure.

Ans. Hydrocarbons

Hydrocarbons are those compounds which are made up of only carbon and hydrogen elements.

Hydrocarbons are regarded as the parent organic compounds since other organic compounds are considered to be derived from them by replacement of one or more hydrogen atoms by other atoms or group of atoms.

Classification of Hydrocarbons

On the basis of structure, hydrocarbons are divided into two main classes:

(i) Open chain or Aliphatic hydrocarbons

These are compounds in which the first and the last carbon are not directly joined to each other. The open chains of carbon may be straight or branched.

Examples;



Straight chain (n-butane)



Branched chain (isobutane)

Types of open chain hydrocarbons

Open chain hydrocarbons have been further subdivided into saturated and unsaturated hydrocarbons.

(a) Saturated hydrocarbons

The compounds in which all the four valencies of carbon atoms are fully satisfied (saturated) by single bonds with other carbon atoms and hydrogen atoms are called saturated hydrocarbons. Saturated hydrocarbons are also called alkanes.

Thus, an alkane is a hydrocarbon in which the carbon atoms are connected by only single covalent bond (There are no double or triple covalent bonds in alkanes).

Examples:

Methane (CH_4), ethane (C_2H_6), propane (C_3H_8) and butane (C_4H_{10}) are all saturated

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