

Chemistry First Semester Review

Key

Introduction to Chemistry

Terms to Define

- qualitative data
- quantitative data
- independent variable
- dependent variable
- control
- constants
- scientific theory
- scientific law
- standard of measurement
- base unit
- derived unit
- accuracy
- precision

Identify the following examples of data as qualitative (QL) or quantitative (QT).

- QL 1. the solution is blue
QL 2. the metal has luster
QN 3. the mass is 105 g
QN 4. the temperature is 12 °C
QN 5. the length is 5.0 mm
QL 6. the liquid conducts electricity

Identify the SI base unit used for each of the following:

| | |
|-----------------------|-----|
| time | s |
| length | m |
| mass | kg |
| temperature | K |
| amount of a substance | mol |

Metric Conversions

| convert | to | |
|-----------|----|-----------------------|
| 0.0032 mg | g | 3.2×10^{-6} |
| 2390 km | m | 2.39×10^6 |
| 0.009 g | mg | 9×10^4 |
| 0.033 Mg | cg | 3.3×10^4 |
| 0.00032 s | ms | 0.32 |
| 3.42 µg | kg | 3.42×10^{-9} |

Temperature Conversions

Convert the following:

- 1) $362\text{ °C} = 635\text{ K}$
 2) $25\text{ °C} = 298\text{ K}$
 3) $300\text{ K} = 27\text{ °C}$
 4) $0\text{ K} = -273\text{ °C}$
 5) $25\text{ K} = -248\text{ °C}$

Experimental Design

You design an experiment to determine which household cleaner best prevents the growth of bacteria. You gather five different household cleaners and some petri dishes with bacteria samples. Determine each of the following for your experiment:

independent variable: type of cleaner
 dependent variable: how well cleaners control bacteria
 control: petri dish w/ bacteria, no cleaner
 constants: temp, strain of bacteria, sunlight, etc

Scientific Notation

Convert the following to scientific or standard notation:

- 1) $8,320,000 = 8.32 \times 10^6$
 2) $0.000937 = 9.37 \times 10^{-4}$
 3) $7,500,000,000 = 7.5 \times 10^9$
 4) $5.22 \times 10^4 = 52,200$
 5) $9.214 \times 10^6 = 9,214,000$
 6) $8.5 \times 10^{-8} = 0.00000085$

Scientific Notation

Perform the following calculations:

- 1) $(2.1 \times 10^5) \times (8.33 \times 10^4) = 1.7 \times 10^{10}$
 2) $(7.50 \times 10^{30}) / (4.212 \times 10^{30}) = 1.78 \times 10^{-3}$
 3) $(4.4 \times 10^6) + (7.2 \times 10^6) = 5.1 \times 10^6$
 4) $(5.29 \times 10^9) - (6.04 \times 10^9) = -5.51 \times 10^9$

Mcps Chemistry Semester Review Guide Answers

Christian G. Meyer



Mcps Chemistry Semester Review Guide Answers:

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properties London dispersion forces magnetic quantum number mass of electron mass spectrometer metallic crystals properties metallic solids metals structure molar volume molecular ions molecular solids molecules moles Moseley law neutron properties non ideal behavior of gases orbital concept partial pressure calculations phase changes energies photons wave number Planck s quantum theory plasma state positive and negative ions pressure units properties of cathode rays covalent crystals properties of crystalline solids properties of positive rays quantum numbers quantum theory relative abundance Rutherford model of atom shapes of orbitals solid iodine structure solids properties solvent extraction spectrometer spin quantum number states of matter stoichiometry sublimation thermometry scales types of solids unit cell van der Waals equation vapor pressure and spectrum

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1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Mcps Chemistry Semester Review Guide Answers
- Chapter 2: Essential Elements of Mcps Chemistry Semester Review Guide Answers
- Chapter 3: Mcps Chemistry Semester Review Guide Answers in Everyday Life
- Chapter 4: Mcps Chemistry Semester Review Guide Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Mcps Chemistry Semester Review Guide Answers. This chapter will explore what Mcps Chemistry Semester Review Guide Answers is, why Mcps Chemistry Semester Review Guide Answers is vital, and how to effectively learn about Mcps Chemistry Semester Review Guide Answers.
3. In chapter 2, this book will delve into the foundational concepts of Mcps Chemistry Semester Review Guide Answers. The second chapter will elucidate the essential principles that must be understood to grasp Mcps Chemistry Semester Review Guide Answers in its entirety.
4. In chapter 3, the author will examine the practical applications of Mcps Chemistry Semester Review Guide Answers in daily life. The third chapter will showcase real-world examples of how Mcps Chemistry Semester Review Guide Answers can be effectively utilized in everyday scenarios.
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6. In chapter 5, this book will draw a conclusion about Mcps Chemistry Semester Review Guide Answers. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Mcps Chemistry Semester Review Guide Answers.

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web planets and their moons the eight planets can be divided into two distinct categories on the basis of their densities mass per unit volume the four inner or terrestrial planets mercury venus earth and mars have rocky compositions and densities greater than 3 grams per cubic cm water has a density of 1 gram per cubic cm in contrast the

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web mar 22 2023 the latest our solar system consists of our star the sun and everything bound to it by gravity the planets mercury venus earth mars jupiter saturn uranus and neptune dwarf planets such as pluto dozens of moons and millions of asteroids comets and meteoroids beyond our own solar system we have discovered thousands

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web mar 15 2016 published march 15 2016 our solar system is made up of a star the sun eight planets 146 moons a bunch of comets asteroids and space rocks ice and several dwarf planets such as pluto the eight planets are mercury venus earth mars jupiter saturn uranus and neptune mercury is closest to the sun neptune is the

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web the solar system s eight planets and its nine most likely dwarf planets are known to be orbited by at least 297 natural satellites or moons at least 19 of them are large enough to be gravitationally rounded of these all are covered by a crust of ice except for earth s moon and jupiter s io several of the largest ones are in hydrostatic equilibrium and would

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web mar 22 2023 the moon was likely formed after a mars sized body collided with earth earth s moon is the fifth largest of the 200 moons orbiting planets in our solar system earth s only natural satellite is simply called the moon because people didn t know other moons existed until galileo galilei discovered four moons orbiting jupiter in 1610

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web the moon is a satellite planet in geophysical terms and among all planetary mass objects of the solar system larger and more massive than all known solar dwarf planets the moon s mass density and surface gravity of about one sixth of earth s at 0 1654 g are rivaled among solar system satellites only by jupiter s moon io

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web there are 171 moons or natural satellites orbiting the planets in our solar system earth mars jupiter saturn uranus and neptune have 1 2 66 62 27 and 13 moons respectively the following is a list of some of the major planetary moons including those of the dwarf planet

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web may 30 2023 uranus and neptune uranus has 27 moons that we know of some of them are half made of ice lastly neptune has 14 named moons one of neptune s moons triton is as big as dwarf planet pluto to learn more about the moons in our solar system visit the nasa solar system exploration moons page article last updated may 30 2023

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web earth s moon and europa one of four planet size moons in orbit around the planet jupiter europa has an almost glassy

surface of ice that scientists say may cover an ocean that is 31 miles 50

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web moons also called satellites come in many shapes sizes and types they are generally solid bodies and few have atmospheres most of the planetary moons probably formed from the discs of gas and dust circulating around planets in the early solar system some moons are large enough for their gravity to cause

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web neptune s moon proteus is on the edge they are shown here to scale with each other two of them are larger than mercury seven are larger than pluto and eris if they were not orbiting planets many of these worlds would be called planets and scientists who study them are called planetary scientists

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web may 23 2023 moons come in many shapes sizes and types a few have atmospheres and even hidden oceans beneath their surfaces most planetary moons probably formed from the discs of gas and dust circulating around planets in the early solar system though some are captured objects that formed elsewhere and fell into orbit around larger

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web callisto is the most heavily cratered object in our solar system four more innermost moons are the the source of the dust

that makes up jupiter s four faint rings the rest of jupiter s moons orbit much farther out from the planet the larger moons may be captured asteroids and the smaller ones are likely fragments from massive collisions

moon national geographic society - Jan 02 2022

web dec 20 2022 a moon is an object that orbits a planet or something else that is not a star besides planets moons can circle dwarf planets large asteroids and other bodies objects that orbit other objects are also called satellites so moons are sometimes called natural satellites people have launched many artificial satellites into orbit around

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web moons in the solar system there are currently 181 known moons in our solar system orbiting the various planets and dwarf planets of the 13 planets and dwarf planets there are four which don t have any moons these are the planets mercury and venus and the dwarf planets ceres and makemake