BOARD QUESTION PAPER: MARCH 2022 CHEMISTRY

Time: 3 Hrs Max. Marks: 70

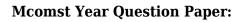
General Instructions:

The question paper is divided into four sections.

	(1)	Section A: Q. No. I contains Ten multiple choice type of questions carrying One mark each. O. No. 2 contains Eight very short answer type of questions carrying One mark each.	
	(2)	Section B: Q. No. 3 to Q. No. 14 are Twelve short answer type of questions carrying Two marks each. (Attempt any Eight).	
	(3)	Section C: Q. No. 15 to Q. No. 26 are Twelve short answer type of questions carrying Three marks each. (Attempt any Eight).	
	(4)	Section D: Q. No. 27 to Q. No. 31 are Five long answer type of questions carrying Four marks each. (Attempt any Three).	
	(5)	Use of log table is allowed. Use of calculator is not allowed.	
	(6)	Figures to the right indicate full marks.	
	(7)	For each multiple choice type of question, it is mandatory to write the correct answer along with its	
		alphabet. e.g. (a)/(b)/(c) etc.	
		narks(s) shall be given, if ONLY the correct answer or the alphabet of the correct answer is written.	
	Only the first attempt will be considered for evaluation.		
		SECTION - A	
Q.1.	Select and write the correct answer for the following multiple choice type of questions: [10]		
	(i)	The co-ordination number of atoms in body centred cubic structure (bcc) is	
		(a) 4 (b) 6 (c) 8 (d) 12	
	(iii)	In calculating osmotic pressure, the concentration of solute is expressed in	
		(a) molarity (b) molality	
		(c) mole fraction (d) percentage mass	
	(iiii)	The enthalpy change for the chemical reaction $H_2O_{(s)}\longrightarrow H_2O_{(t)}$ is called enthalpy of	
		(a) vapourisation (b) fusion	
		(c) combustion (d) sublimation	
	(iv)	Which of the following transition element shows maximum oxidation state?	
	5335	(a) Sc (b) Fe (c) Mn (d) V	
	(v)	The correct formula for the complex compound, sodium hexacyanoferrate (III) is	
	20004	(a) Na [Fe(CN) ₆] (b) Na ₂ [Fe(CN) ₆]	
		(c) Na ₃ [Fe(CN) ₆] (d) Na ₄ [Fe(CN) ₆]	

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Table of Contents Mcomst Year Question Paper

- 1. Understanding the eBook Mcomst Year Question Paper
 - The Rise of Digital Reading Mcomst Year Question Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Mcomst Year Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcomst Year Question Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mcomst Year Question Paper
 - Personalized Recommendations
 - Mcomst Year Question Paper User Reviews and Ratings
 - Mcomst Year Question Paper and Bestseller Lists
- 5. Accessing Mcomst Year Question Paper Free and Paid eBooks

- Mcomst Year Question Paper Public Domain eBooks
- Mcomst Year Question Paper eBook Subscription Services
- Mcomst Year Question Paper Budget-Friendly Options
- 6. Navigating Mcomst Year Question Paper eBook Formats
 - o ePub, PDF, MOBI, and More
 - Mcomst Year Question Paper Compatibility with Devices
 - Mcomst Year Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mcomst Year Question Paper
 - Highlighting and Note-Taking Mcomst Year Question Paper
 - Interactive Elements Mcomst Year Question Paper
- 8. Staying Engaged with Mcomst Year Question Paper
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcomst Year Question Paper
- 9. Balancing eBooks and Physical Books Mcomst Year Question Paper
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Mcomst Year Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mcomst Year Question Paper
 - Setting Reading Goals Mcomst Year Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcomst Year Question Paper
 - Fact-Checking eBook Content of Mcomst Year Question Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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