General Chemistry 4th Edition

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Chlorine can be prepared in the laboratory by the reaction of manganese dioxide with hydrochloric acid, HCl(aq), as described by the chemical equation

$$\mathrm{MnO}_2(x) + 4 \mathrm{HCl}(\mathrm{aq}) \longrightarrow \mathrm{MnCl}_2(\mathrm{aq}) + 2 \mathrm{H}_2\mathrm{O}(t) + \mathrm{Cl}_2(g)$$

How much MnO₂(s) should be added to excess HCl(aq) to obtain 115 mL Cl₂(g) at 25 °C and 775 Torr?

mass of MiO₂: 13

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Carole H. McQuarrie

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