

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Chapter 16 Review Answers

Yi-Tong Ma



Modern Chemistry Chapter 16 Review Answers:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Independent and Weekly Review**, 1914 **Fundamentals of Fragrance Chemistry** Charles S. Sell, 2019-08-12

Comprehensively teaches all of the fundamentals of fragrance chemistry Ernest Beaux the perfumer who created Chanel No 5 said One has to rely on chemists to find new aroma chemicals creating new original notes In perfumery the future lies primarily in the hands of chemists This book provides chemists and chemists to be with everything they need to know in order to create welcome new fragrances for the world to enjoy It offers a simplified introduction into organic chemistry including separation techniques and analytical methodologies discusses the structure of perfume creation with respect to the many reactive ingredients in consumer products and shows how to formulate effective and long lasting scents Fundamentals of Fragrance Chemistry starts by covering the structure of matter in order to show how its building blocks are held together It continues with chapters that look at hydrocarbons and heteroatoms A description of the three states of matter and how each can be converted into another is offered next followed by coverage of separation and purification of materials Other chapters examine acid base reactions oxidation and reduction reactions perfume structure the mechanism of olfaction natural and synthetic fragrance ingredients and much more Concentrates on aspects of organic chemistry which are of particular importance to the fragrance industry Offers non chemists a simplified yet complete introduction to organic chemistry from separation techniques and analytical methodologies to the structure of perfume creation Provides innovative perfumers with a framework to formulate stable fragrances from the myriad of active ingredients available Looks at future trends in the industry and addresses concerns about sustainability and quality management Fundamentals of Fragrance Chemistry is an ideal resource for students who are new to the subject as well as for chemists and perfumers already working in this fragrant field of science

The Handy Chemistry Answer Book Justin P. Lomont, Ian C. Stewart, 2013-10-01 Don't be mixed up about chemistry Simplify the complex chemical reactions that take place everywhere in our lives with this engaging easy to follow question and answer guide Where would we be without atoms and compounds Gas liquids solids and plasma Acids and bases Bonds and reactions Matter and energy The Handy Chemistry Answer Book covers the building blocks of life and the universe The secret life of atoms how polar bears aren't actually white why oil and water don't mix and much much more are revealed and explained This informative guide covers the basics of chemistry history atomic structures chemical bonds and reactions organic and inorganic chemistry to more advanced material nuclear chemistry biochemistry physical and theoretical chemistry by answering nearly 1 000 common chemistry questions including What causes lightning How does photosynthesis work What are hard and soft Lewis acids and bases What makes a fabric waterproof What are the twelve principles of green chemistry When did alchemists finally abandon trying to make gold What is Le Chatelier's principle What do the different octane ratings mean at the gas pump What is genetic engineering Why is calcium important for strong bones

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Chemistry Henry Dorin,1987 **Freud's Answer** Martin Wain,1998-10 Now that authority in the form of the king the church and the family had lost its power the new liberal democracies needed new means of control *Academy, with which are Incorporated Literature and the English Review* ,1882 *Educational Times* ,1909 **The Bookseller** ,1913 Official organ of the book trade of the United Kingdom **Chemical News and Journal of Physical Science** ,1913

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