

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- | | | | |
|-------|----------|---------------------------|--|
| C.2.2 | <u>a</u> | 1. cation | a. a positively charged ion |
| C.2.2 | <u>B</u> | 2. electronegativity | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled |
| C.2.5 | <u>J</u> | 3. inner transition metal | c. the energy required to remove an electron from an atom |
| C.2.6 | <u>I</u> | 4. alkali earth metal | d. good conductor of heat and electric current |
| C.2.6 | <u>h</u> | 5. noble gas | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons |
| C.2.5 | <u>b</u> | 6. representative element | f. any nonmetal in group 7A |
| C.2.2 | <u>d</u> | 7. metal | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>f</u> | 8. halogen | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled |
| C.2.5 | <u>e</u> | 9. transition metal | i. an element in group 2A |
| C.2.2 | <u>c</u> | 10. ionization energy | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons |

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- | | | | | |
|-------|----------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u>b</u> | 11. The alkali metals do not include | a. Li | c. Na |
| | | | b. Ca | d. Rb |
| C.2.6 | <u>c</u> | 12. The element iodine, I, is a | a. period 5 alkali metal | c. period 5 halogen |
| | | | b. period 4 halogen | d. period 5 transition metal |

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Organic Chemistry Workbook** Pierre Vogel, Kendall N. Houk, 2019-08-20 Provides references and answers to every question presented in the primary Organic Chemistry textbook Successfully achieving chemical reactions in organic chemistry requires a solid background in physical chemistry Knowledge of chemical equilibria thermodynamics reaction rates reaction mechanisms and molecular orbital theory is essential for students chemists and chemical engineers The Organic Chemistry presents the tools and models required to understand organic synthesis and enables the efficient planning of chemical reactions This volume Organic Chemistry Theory Reactivity and Mechanisms in Modern Synthesis Workbook complements the primary textbook supplying the complete calculated solutions to more than 800 questions on topics such as thermochemistry pericyclic reactions organic photochemistry catalytic reactions and more This companion workbook is indispensable for those seeking clear in depth instruction on this challenging subject Written by prominent experts in the field of organic chemistry this book Works side by side with the primary Organic Chemistry textbook Includes chapter introductions and re stated questions to enhance efficiency Features clear illustrations tables and figures Strengthens reader s comprehension of key areas of knowledge Organic Chemistry Theory Reactivity and Mechanisms in Modern Synthesis Workbook is a must have resource for anyone using the primary textbook **Principles of Modern Chemistry** David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom **Modern Chemistry** Raymond E. Davis, 1999 2000 2005 State Textbook Adoption Rowan Salisbury **Organic Synthesis** Michael Smith, 2011-07-12 The first two chapters provide an introduction to functional groups these are followed by chapters reviewing basic organic transformations e g oxidation reduction The book then looks at carbon carbon bond formation reactions and ways to disconnect a bigger molecule into simpler building blocks Most chapters include an extensive list of questions to test the reader s understanding There is also a new chapter outlining full retrosynthetic analyses of complex molecules which highlights common problems made by scientists National Geographic Answer Book National Geographic, 2011 Covers everything from earth sciences to

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