

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Chapter 9 Answers

Eric V. Anslyn, Dennis A. Dougherty



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Solutions to Problems, Principles of Modern Organic Chemistry James Cason, 1967 **Solutions Manual for Modern Organic Synthesis: An Introduction** Michael Nantz, Hasan Palandoken, George S. Zweifel, 2006-08-18 This supplement includes the end of chapter problems from the main text detailed solution sets and an extra section of similar problems for grad students to study Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

Sustainable Solutions for Modern Economies Rainer Höfer, 2009-11-20 Limited supplies of fossil fuels and concerns about global warming have created a strong desire to solve the resource issue in the age beyond petroleum This reference book from the Green Chemistry Series contains the essential areas of green chemistry and sustainability in modern economies It is the first book to outline the contribution of chemistry and of renewable chemical or biological resources to the sustainability concept and to the potential resolution of the world's energy problems It describes the current status of technical research and industrial application as well as the potential of biomass as a renewable resource for energy generation in power stations as alternative fuels and for various uses in chemistry It outlines the historical routes of the sustainability concept and specifies sustainability in metrics facts and figures The book is written by European experts from academia industry and investment banking who are world leaders in research and technology regarding sustainability alternative energies and renewable resources The sustainability aspects covered include consumer behaviour and demands lifestyles and mega trends and their impact on innovation in the industry consumer industry requirements and their impact on suppliers emerging paradigm changes in raw material demand availability sourcing and logistics the contribution of the industry to restore the life support systems of the Earth socially responsible banking and investment sustainability metrics The book highlights the potential of the different forms of renewable raw materials including natural fats and oils plant based biologically active ingredients industrial starch sucrose natural rubber wood natural fibres It also covers the actual status of biomass usage for green energy generation green transportation green chemistry and sustainable nutrition and consumer goods and it depicts the potentials of green solvents and white biotechnology for modern synthesis and manufacturing technologies The book is aimed at technical and marketing people in industry universities and institutions as well as readers in administrations and NGOs The book will also be of value to the worldwide public interested in sustainability issues and strategies as well as others interested in the practical means that are being used to reduce the environmental impact of chemical processes and products to further eco efficiency and to advance the utilization of renewable resources

Electrochemistry in Nonaqueous Solutions Kosuke Izutsu, 2009-09-22 An excellent resource for all graduate students and researchers using electrochemical techniques After introducing the reader to the fundamentals the book focuses on the latest developments in the techniques and applications in this field This second edition contains new material on environmentally friendly solvents such as room temperature ionic liquids **Modern Physical Organic Chemistry** Eric V.

Anslyn, Dennis A. Dougherty, 2006 Making explicit the connections between physical organic chemistry and critical fields such as organometallic chemistry materials chemistry bioorganic chemistry and biochemistry this book escorts the reader into an area that has been thoroughly updated in recent times **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** McMurry, Susan McMurry, 1990 *Student's Guide to Chemistry; a Modern Introduction* David Brooks, 1974 **Catechism of Modern Elementary Chemistry Or Solutions of the Questions Set at the London University Matriculation Examinations 1844-82** E. W. V. Volckxsom, 1882 **Introduction to Modern Chemistry** Edward Florian Neuzil, 1968 *Supplement for Modern Organic Chemistry* John D. Roberts, Marjorie C. Caserio, 1967 **Study Guide and Solutions Manual for Organic Chemistry** Susan McMurry, 1992 John McMurry's best selling text presents organic chemistry in a new edition that is up to date beautifully written visually striking and pedagogically sound Described by many of its users as an eminently teachable text McMurry sets the standard in the field The writing style has received almost universal acclaim from its users McMurry introduces new concepts only as needed and immediately illustrates them with concrete examples And wherever possible he ties material together with brief reviews overviews and reaction summaries The result is a text that helps students mentally organize the material a text that helps them understand concepts not just memorize facts and a text that helps them make sense of the voluminous amount of material they encounter in the study of organic chemistry McMurry uses a simple but important polar reaction the addition of HBr to an alkene as the lead off reaction to illustrate the general principles of organic reactions Users of former editions found this an excellent choice because of its relative simplicity no prior knowledge of chirality or kinetics is required and its importance as a polar reaction on a common functional group that offers students the key to understanding hundreds of thousands of ionic reactions By selecting this particular model McMurry is able to offer an unusually early presentation of organic reactions Introduction to Modern Inorganic Chemistry Joseph William Mellor, 1914 *Solutions Guide for Introductory Chemistry* James F. Hall, 1996 *Principles of Classical and Modern Physics* Mr. Ajeet Singh, 2018-05-16 EduGorilla has its own publishing wing producing exam prep books trade books etc *Modern University Chemistry, Custom Pub* Norbert T. Porile, 1993 **Study and Problem Solving Guide to Accompany Principles of Modern Chemistry, Oxtoby/Nachtrieb** Wade A. Freeman, 1986 **The Independent**, 1914 *The Independent* William Livingston, 1913 *Solutions Guide, Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry, Fourth Edition, Zumdahl* James F. Hall, 2000

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