

AP Chemistry Pacing Guide

Unit 4: Chemical Reactions - Introduction for Reactions - Net Ionic Equations - Representations of Reactions - Physical and Chemical Changes - Stoichiometry - Introduction to Titration - Types of Chemical Reactions - Introduction to Acid-Base Reactions - Oxidation-Reduction (Redox) Reactions	BIG IDEA 1: Scale, Proportion, and Quantity BIG IDEA 3: Transformations	- I can identify evidence of chemical and physical changes in matter. - I can represent changes in matter with a balanced chemical or net ionic equation: a. For physical changes; b. For given information about the identity of the reactants and/or product. - For ions in a given chemical reaction I can represent a given chemical reaction or physical process with a consistent particulate model. - I can explain the relationship between macroscopic characteristics and bond interactions for: a. Chemical processes; b. Physical processes. - I can explain changes in the amounts of reactants and products based on the balanced reaction equation for a chemical process. - I can identify the equivalence point in a titration based on the amounts of the titrant and analyte, assuming the titration reaction goes to completion. - I can identify a reaction as acid-base, oxidation-reduction, or precipitation. - I can identify species as Brønsted-Lowry acids, bases, and/or conjugate acid-base pairs, based on proton-transfer involving those species. - I can represent a balanced redox reaction equation using half-reactions.	10 days
Unit 5: Kinetics - Reaction Rates - Introduction to Rate Law - Concentration Changes Over Time - Elementary Reactions - Collision Model - Reaction Energy Profile - Introduction to Reaction - Mechanisms Reaction Mechanism and Rate Law - Steady-State	BIG IDEA 3: Transformations BIG IDEA 6: Energy	- I can explain the relationship between the rate of a chemical reaction and experimental parameters. - I can represent experimental data with a consistent rate law expression. - I can identify the rate law expression of a chemical reaction using data that show how the concentrations of reaction species change over time. - I can represent an elementary reaction as a rate law expression using stoichiometry. - I can explain the relationship between the rate of an elementary reaction and the frequency, energy, and orientation of molecular collisions. - I can represent the activation energy and overall energy change in an elementary reaction using a reaction energy profile. - I can identify the components of a reaction mechanism. - I can identify the rate law for a reaction from a mechanism in which the first step is rate limiting. - I can identify the rate law for a reaction from a mechanism in which the	10 days

Modern Chemistry Pacing Guide

Lei Huang



Modern Chemistry Pacing Guide:

Modern Chemistry Pacing Guide Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the ability of words has been evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Modern Chemistry Pacing Guide**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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