

Multiple Choice Reaction Mechanisms

23. Given the following reaction mechanism, what is the equation for the overall reaction?



ANS: A

11. Which step below is the rate-determining step?



fast



very fast



slow



fast

A) 1

B) 2

C) 3

D) 4

ANS: C

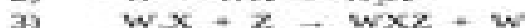
12. Which of the following rate law expressions would support the mechanism below?



fast



very fast



slow



fast



ANS: A

11. Consider the following reaction mechanism:



SLOW



FAST

The overall reaction is:



How might the rate of the forward reaction be increased?

A) increase [A]

B) increase [B]

C) decrease temperature

D) both A and B

ANS: A

4. Consider the following reaction mechanism:



In this reaction, H_2 is a:

Modern Chemistry Reaction Energy Multiple Choice Answers

YCT Expert Team



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