

**Stoichiometry Review**

**Directions:** Each of the problems below reviews a different type of stoichiometry. Be sure to read the problem carefully to know what it is asking for.

**Mass-to-Mass Problems**

1. How many grams of potassium chloride, KCl, are produced during the decomposition of 100.0 grams of potassium chlorate,  $\text{KClO}_3$ ?



$$\frac{100.0 \text{ g KClO}_3}{1} \times \frac{1 \text{ mol KClO}_3}{122.545 \text{ g KClO}_3} \times \frac{1 \text{ mol KCl}}{1 \text{ mol KClO}_3} \times \frac{74.548 \text{ g KCl}}{1 \text{ mol KCl}} = \underline{60.83 \text{ g KCl}}$$

2. Find the number of grams of HCl needed to react completely with 0.50 moles of magnesium.



$$\frac{0.5 \text{ mol Mg}}{1} \times \frac{1 \text{ mol HCl}}{1 \text{ mol Mg}} \times \frac{36.458 \text{ g HCl}}{1 \text{ mol HCl}} = \underline{18.23 \text{ g HCl}}$$

3. Find the mass in grams of hydrogen gas produced when 14.0 moles of HCl are added to an excess amount of magnesium.



$$\frac{14 \text{ mol HCl}}{1} \times \frac{1 \text{ mol H}_2}{2 \text{ mol HCl}} \times \frac{2.016 \text{ g H}_2}{1 \text{ mol H}_2} = \underline{14.112 \text{ g H}_2}$$

4. If 0.75 moles of ammonia are needed, how many grams of nitrogen will be consumed?



$$\frac{0.75 \text{ mol NH}_3}{1} \times \frac{1 \text{ mol N}_2}{2 \text{ mol NH}_3} \times \frac{28.014 \text{ g N}_2}{1 \text{ mol N}_2} = \underline{10.51 \text{ g N}_2}$$

# Modern Chemistry Stoichiometry Review Answers

**Yicheng Fang**



## **Modern Chemistry Stoichiometry Review Answers:**

This book delves into Modern Chemistry Stoichiometry Review Answers. Modern Chemistry Stoichiometry Review Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Modern Chemistry Stoichiometry Review Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Modern Chemistry Stoichiometry Review Answers
- Chapter 2: Essential Elements of Modern Chemistry Stoichiometry Review Answers
- Chapter 3: Modern Chemistry Stoichiometry Review Answers in Everyday Life
- Chapter 4: Modern Chemistry Stoichiometry Review Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Modern Chemistry Stoichiometry Review Answers. The first chapter will explore what Modern Chemistry Stoichiometry Review Answers is, why Modern Chemistry Stoichiometry Review Answers is vital, and how to effectively learn about Modern Chemistry Stoichiometry Review Answers.

3. In chapter 2, the author will delve into the foundational concepts of Modern Chemistry Stoichiometry Review Answers. This chapter will elucidate the essential principles that need to be understood to grasp Modern Chemistry Stoichiometry Review Answers in its entirety.

4. In chapter 3, this book will examine the practical applications of Modern Chemistry Stoichiometry Review Answers in daily life. The third chapter will showcase real-world examples of how Modern Chemistry Stoichiometry Review Answers can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Modern Chemistry Stoichiometry Review Answers in specific contexts. This chapter will explore how Modern Chemistry Stoichiometry Review Answers is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Modern Chemistry Stoichiometry Review Answers. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Modern Chemistry Stoichiometry Review Answers.

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