

CHAPTER 6 REVIEW*Chemical Bonding***MIXED REVIEW**

SHORT ANSWER Answer the following questions in the space provided.

1. Name the type of energy that is a measure of strength for each of the following types of bonds:

_____ a. ionic bond
_____ b. covalent bond
_____ c. metallic bond

2. Use the electronegativity values shown in Figure 5-20, on page 151 of the text, to determine whether each of the following bonds is nonpolar covalent, polar covalent, or ionic.

_____ a. H—F	_____ d. H—H
_____ b. Na—Cl	_____ e. H—C
_____ c. H—O	_____ f. H—N

3. How is a hydrogen bond different from an ionic or covalent bond?

4. H_2S and H_2O have similar structures and their central atoms belong to the same group. Yet H_2S is a gas at room temperature and H_2O is a liquid. Use bonding principles to explain why this is true.

Modern Chemistry Chapter 6 Review Chemical Bonding

Norbert T. Porile



Modern Chemistry Chapter 6 Review Chemical Bonding:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Modern Chemistry* Nicholas D. Tzimopoulos, 1993 *Instructor's Manual and Test Bank to Accompany Basic Concepts of Chemistry* Malone, 2000-04-18

Chemistry George Moriber, 1980 A textbook introducing the sources production and uses of chemicals **From Classical to Modern Chemistry** Peter J. T. Morris, Peter John Turnbull Morris, Royal Society of Chemistry (Great Britain), 2002 Most chemists today have either taken part in or been affected by the chemical revolution that has taken place over the course of the last century Developments in instrumentation have changed not just what chemists do but also how they think about chemistry New and exciting areas of previously inaccessible research have been opened up as a direct result of this revolution This is the first book to examine this instrumental revolution and goes on to assess the impact on chemical practice in areas ranging from organic chemistry and biochemistry to environmental analysis and process control thus demonstrating how fundamental and extensive are the changes that have occurred With contributions from internationally recognised specialists this lavishly illustrated book provides a focal point for any historian of chemistry or chemist with an interest in this fascinating topic This book is published in association with the Science Museum London UK and the Chemical Heritage Foundation Philadelphia Modern Chemistry H. Clark Metcalfe, Joseph F. Castka, 1966 *Solutions Guide for Introductory Chemistry* James F. Hall, 1996 Modern University Chemistry, Custom Pub Norbert T. Porile, 1993

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