

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.** C In a Stock name such as iron(III) sulfate, the roman numeral tells us \_\_\_\_\_.
  - (a) how many atoms of Fe are in one formula unit
  - (b) how many sulfate ions can be attached to the iron atom
  - (c) the charge on each Fe ion
  - (d) the total positive charge of the formula unit
- 2.** C The result of changing a subscript in a correctly written chemical formula is to \_\_\_\_\_.
  - (a) change the number of moles represented by the formula
  - (b) change the charges on the other ions in the compound
  - (c) change the formula so that it no longer represents the compound it previously represented
  - (d) have no effect on the formula
- 3.** The explosive TNT has the molecular formula  $C_7H_5(NO_2)_3$ .
 

<u>4 elements</u>	<b>a.</b> How many elements make up this compound?
<u>6 oxygen atoms</u>	<b>b.</b> How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$ ?
<u>21 atoms</u>	<b>c.</b> How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$ ?
<u><math>4.2 \times 10^{24}</math> atoms</u>	<b>d.</b> How many atoms are present in a sample of $2 \times 10^{23}$ molecules of $C_7H_5(NO_2)_3$ ?
- 4.** How many atoms are present in each of these formula units?
 

<u>11 atoms</u>	<b>a.</b> $Ca(HCO_3)_2$
<u>45 atoms</u>	<b>b.</b> $C_{12}H_{22}O_{11}$
<u>10 atoms</u>	<b>c.</b> $Fe(ClO_2)_3$
<u>9 atoms</u>	<b>d.</b> $Fe(ClO_3)_3$
- 5.**

<u><math>N_2O_5</math></u>	<b>a.</b> What is the formula for the compound dinitrogen pentoxide?
<u>carbon(IV) sulfide</u>	<b>b.</b> What is the Stock name for the covalent compound $CS_2$ ?
<u><math>H_2SO_3</math></u>	<b>c.</b> What is the formula for sulfurous acid?
<u>phosphoric acid</u>	<b>d.</b> What is the name for the acid $H_3PO_4$ ?

# Modern Chemistry Chapter 7 Review Chemical Formulas And Compounds Answers

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Stanitski, Peter C. Jurs**



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