

## Molarity Worksheet

Name: \_\_\_\_\_

1. What does molarity mean?

$$\frac{\text{Number of moles of solute}}{\text{1 liter solution}}$$

2. What is the molarity of a solution that contains 4.53 moles of lithium nitrate in 2.85 liters of solution?

$$\frac{4.53 \text{ mol LiNO}_3}{2.85 \text{ L soln}} = 1.59 \text{ M LiNO}_3$$

3. What is the molarity of a solution that contains 0.00372 moles hydrochloric acid in  $2.39 \times 10^{-2}$  liters of solution?

$$\frac{0.00372 \text{ mol HCl}}{2.39 \times 10^{-2} \text{ L soln}} = 0.156 \text{ M HCl}$$

4. A flask contains 85.5 g  $\text{C}_{12}\text{H}_{22}\text{O}_{11}$  (sucrose) in 1.00 liters of solution. What is the molarity?

$$\frac{85.5 \text{ g sucrose} \times 1 \text{ mol sucrose}}{1.00 \text{ L soln} \quad 342.34 \text{ g sucrose}} = 0.250 \text{ M sucrose}$$

5. A beaker contains 214.2 grams osmium (III) fluoride in 0.0673 liters of solution. What is the molarity?

$$\frac{214.2 \text{ g OsF}_3 \times 1 \text{ mol OsF}_3}{0.0673 \text{ L soln} \quad 247.23 \text{ g OsF}_3} = 12.9 \text{ M OsF}_3$$

6. Calculate the molarity if a flask contains 1.54 moles potassium sulfate in 125 ml of solution.

$$\frac{1.54 \text{ mol K}_2\text{SO}_4}{0.125 \text{ L soln}} = 12.3 \text{ M K}_2\text{SO}_4$$

7. A chalice contains 36.45 grams ammonium chlorite in 2.36 liters of solution - calculate the molarity.

$$\frac{36.45 \text{ g NH}_4\text{ClO}_2 \times 1 \text{ mol NH}_4\text{ClO}_2}{2.36 \text{ L soln} \quad 85.50 \text{ g NH}_4\text{ClO}_2} = 0.181 \text{ M NH}_4\text{ClO}_2$$

8. What is the molarity of a solution that contains 14.92 grams magnesium oxalate in 3.65 ml of solution?

$$\frac{14.92 \text{ g MgC}_2\text{O}_4 \times 1 \text{ mol MgC}_2\text{O}_4}{0.00365 \text{ L soln} \quad 112.32 \text{ g MgC}_2\text{O}_4} = 36.4 \text{ mol MgC}_2\text{O}_4$$

# Molarity Practice Unit Lesson Key

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**Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation

*Chemistry for the Biosciences* Jonathan Crowe, Tony Bradshaw, 2014 Leading students through the essential concepts that are central to understanding biological systems this text uses everyday examples and analogies to build their confidence in an often daunting subject By focusing on the key themes that unify the subject it shows how integral chemistry is to the biosciences

Foundations of College Chemistry Morris Hein, Susan Arena, 2013-01-01 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years Foundations of College Chemistry Alternate 14th Edition has helped readers master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

**Holt Chemistry** Ralph Thomas Myers, 2004

**The Software Encyclopedia 2001**, 2001

Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics - E-Book Carl A. Burtis, David E. Bruns, 2014-02-26 A condensed easier to understand student version of the acclaimed Tietz Textbook of Clinical Chemistry and Molecular Diagnostics Tietz Fundamentals of Clinical Chemistry and Molecular Diagnostics 7th Edition uses a laboratory perspective in providing the clinical chemistry fundamentals you need to work in a real world clinical lab Coverage ranges from laboratory principles to analytical techniques and instrumentation analytes pathophysiology and more New content keeps you current with the latest developments in molecular diagnostics From highly respected clinical chemistry experts Carl Burtis and David Bruns this textbook shows how to select and perform diagnostic lab tests and accurately evaluate results Authoritative respected author team consists of two well known experts in the clinical chemistry world Coverage of analytical techniques and instrumentation includes optical techniques electrochemistry electrophoresis chromatography mass spectrometry enzymology immunochemical techniques microchips automation and point of care testing Learning objectives begin each chapter providing measurable outcomes to achieve after completing the material Key words are listed and defined at the

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*Health Education* Karen Glanz, Frances Marcus Lewis, Barbara K. Rimer, 1997 In this completely revised and updated edition of *Health Behavior and Health Education* the contributors offer the most current information on developments in health behavior in theory research and practice Drawing from such fields as cognitive and organization psychology marketing and communications the authors offer understanding about the diverse factors affecting health behavior The editors were the recipients of the 1992 Mayhew Derryberry Award for outstanding contribution to health education theory and research

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