Stoichiometry: Mass - Mole Worksheet

Based on the following chemical equation, please answer the following questions!

Fe (SO₄), + 6 NaCl --> 2 FeCl, + 3 Na₂SO₄

Calculate the amount of moles of NaCl needed to produce 75 g of FeCl,.

9 FeCl,	1 mole FeCl,	mole NaCl	
	9 FeCl,	mole FeCl	

Calculate the amount of Fe, (SO,), in grams produce 215 g Na, SO,.

9 Na ₃ SO ₄	1 mole Na SO	mole Fe ₂ (SO ₄),	9 Fe ₂ (SO ₄),
	9 Na ₂ SO ₄	mole Na,SO,	1 mole Fe ₂ (SO ₄),

Calculate the amount in grams of FeCl, produced with 95g of Fe (SO₄),

9 Fe ₂ (SO ₄),	mole Fe2(SO4)	mole FeCl,	9 FeCl,
	9 Fe,(SO,),	mole Fe(SO ₄),	1 mole FeCl,

Calculate the amount of moles of Fe₂(SO₄) needed to react with 123.5 g of NaCl

9 NaCl	1 mole NaCl	mole Fe(SO _e),	
	9 NaCl	mole NaCl	

Based on the following chemical equation, please answer the following questions!

9 Ca(HCO _y) ₂	1 mole Ca(HCO ₅),	mole CaCO,
	9 Ca(HCO,),	mole Ca(HCO,),

Mole And Stoichiometry Practice Answer Key

Disha Experts

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