Mole to Mole Stoichiometry Worksheet

In the reaction below, how many moles of aluminum will be used when reacted with 1.33 moles of oxygen?
 □ Al + 3 O₂
 □ Al₂O₃

2) How many moles of hydrogen will be produced when reacted with 0.0204 moles of sodium in the reaction?

2 Na + 2 H₂O ⇒ 2 NaOH + | H₂

Using the equation
 ⇒ Al(NO₃)₃ + 3 MgBr₂ ⇒ 3 Mg(NO₃)₂ + ⇒ AlBr₃
 a. Find the number of moles Al(NO₃)₃ used to produce 2.3 moles of Mg(NO₃)₂

b. Find the number of moles Al(NO₃)s used to produce 0.78 moles of AlBra

c. Find the number of moles MgBr2 that reacts with 1.05 moles of Al(NO₃)1

4) In the reaction below, how many moles of carbon dioxide are produced if 19.65 moles of water are produced?

∠ C₂H₅ + 7O₂ ⇒ 4 CO₂ + ∠ H₂O

Mole Mole And Mass Mass Stoichiometry Worksheet

TD Snyder

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