

# Molecular Polarity

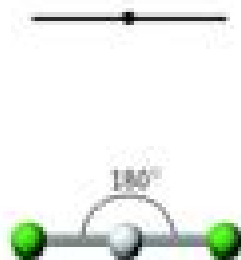
Molecules will be polar if

a) bonds are polar

AND

b) the molecule is NOT "symmetric"

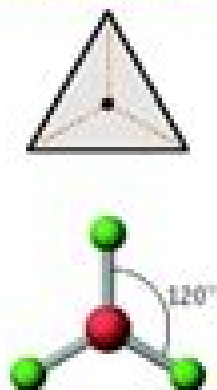
Linear



$AX_2$

Example:  $\text{BeF}_2$

Trigonal-planar



$AX_3$

Example:  $\text{BF}_3$

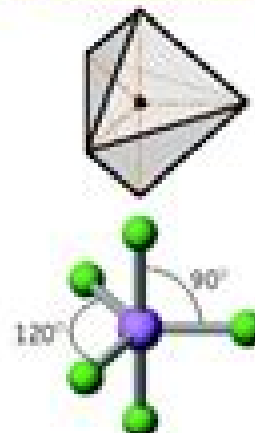
Tetrahedral



$AX_4$

Example:  $\text{CF}_4$

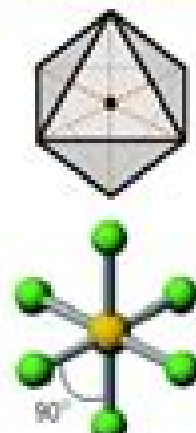
Trigonal-bipyramidal



$AX_5$

Example:  $\text{PF}_5$

Octahedral



$AX_6$

Example:  $\text{SF}_6$

All above are **NOT** polar

# Molecule Polarity Pogil

**Juliette Lantz, Renée Cole, The POGIL  
Project**



## **Molecule Polarity Pogil:**

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