

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20. \\ \hline & + & \\ & 63.62 & \rightarrow 64 \end{array}$$

Modern Chemistry Chapter 5 Section Answers

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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom **Operation of Wastewater Treatment Plants**, 2004 *The Jahn-Teller Effect and Vibronic Interactions in Modern Chemistry* Isaac Bersuker, 2013-06-29 The first half of the title of this book may delude the uninitiated reader The term Jahn Teller effect taken literally refers to a special effect inherent in particular molecular systems Actually this term implies a new approach to the general problem of correlations between the structure and properties of any molecular polyatomic system including solids Just such a new approach or concept in some sense a new outlook or even a new way of thinking which leads not to one special effect but to a series of different effects and laws is embodied in the many 4000 studies devoted to the investigation and application of the Jahn Teller effect The term vibronic interactions seems to be most appropriate to the new concept and this explains the origin of the second half of the title The primary objective of this book is to present a systematic development of the concept of vibronic interactions and its applications and to illustrate its possibilities and significance in modern chemistry In the first three chapters covering about one third of the book the theoretical background of the vibronic concept and Jahn Teller effect is given The basic ideas are illustrated fully although a comprehensive presentation of the theory with all related mathematical deductions is beyond the scope of this book In the last three chapters the applications of theory to spectroscopy stereochemistry and crystal chemistry reactivity and catalysis are illustrated by a series of effects and laws *Modern NMR Techniques for Chemistry Research* A.E. Derome, 2013-10-22 Presents an introduction to modern NMR methods at a level suited to organic and inorganic chemists engaged in the solution of structural and mechanistic problems The book assumes familiarity only with the simple use of proton and carbon spectra as sources of structural information and describes the advantages of pulse and Fourier transform spectroscopy which form the basis of all modern NMR experiments Discussion of key experiments is illustrated by numerous examples of the solutions to real problems The emphasis throughout is on the practical side of NMR and the book will be of great use to chemists engaged in both academic

and industrial research who wish to realise the full possibilities of the new wave NMR **Structure Elucidation by Modern NMR** H. Duddeck, W. Dietrich, 2012-12-06 For several years we have been organizing seminars and workshops on the application of modern one and two dimensional NMR methods at the faculty of chemistry in the Ruhr University Bochum FRG and elsewhere addressing researchers and graduate students who work in the field of organic and natural products chemistry In 1987 we wrote a workbook *Strukturaufklärung mit moderner NMR Spektroskopie* Steinkopff Darmstadt FRG 1988 in German language based on our experience in these courses Many of the exercises described therein have been used in such courses and some of them have been shaped by the participants to a great extent The response of readers and discussions with colleagues from many countries encouraged us to produce an English translation in order to make the book accessible to a wider audience Moreover the content has been increased from 20 exercise examples in the German to 23 in the English version This book could not have been written in the present form without the help of a number of colleagues and therefore we acknowledge gratefully the generous supply of samples from and useful discussions with B Abegaz Addis Ababa Ethiopia U H Brinker Bingham New York USA E **Modern Biopharmaceuticals, 4 Volume Set** Jörg Knäblein, 2005-10-28 The biopharmaceutical market has come along way since 1982 when the first biopharmaceutical product recombinant human insulin was launched Over 120 such products are currently being marketed around the world including nine blockbuster drugs The global market for biopharmaceuticals which is currently valued at US 41 billion has been growing at an impressive compound annual growth rate of 21% over the previous five years With over one third of all pipeline products in active development are biopharmaceuticals this segment is set to continue outperforming the total pharmaceutical market and could easily reach US 100 billion by the end of this decade *The Independent* ,1914
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