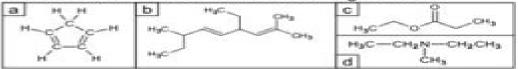
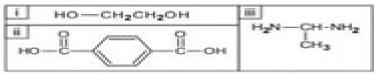
Organic chemistry review

- 1. Draw structures for the following compounds:
 - a) 2-methylbutane
 - b) 3,3-dichlorohexane
 - c) 4-ethyl-3,3,4-trimethyldecane
 - d) ethanamide
 - e) cis-1,3-dimethylcyclohexane

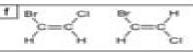
- f) 2-methyl-4-octanol
- g) propanal
- h) methyl ethanoate
- i) methyl propyl ether
- j) 2-pentanone
- Write the correct name for each of the following structures:



- Draw the polymers that result from:
 - a) the reaction of i) and ii)
 - b) the reaction of ii) and iii)
 - c) the addition of ethene molecules
 - d) the addition of 3-hexene molecules



- 4. What is the difference between a saturated and unsaturated hydrocarbon? Which is benzene?
- Identify each pair as structural isomers, geometric isomers, or neither
 - a) cyclopentane, pentane
 - b) 1,1-dichloroethene, trans-1,2-dichloroethene
 - c) hexanoic acid, propyl propanoate
 - d) cis-1,2-dichlorocyclopentane, trans-1,2-dichlorocyclopentane
 - e) cis-1,2-dichlorocyclopentane, trans-1,3-dichlorocyclopentane



- C₂H₆O exists as two structural isomers. Draw each isomer and name each. Predict the relative boiling point and solubility in water of the two isomers. Explain.
- 7. Indicate if the functional group contains a cabonyl, carboxyl or hydroxyl group:

	alcohol	ketone	ester	carboxylic acid
carboxyl group				
carbonyl group				
hydroxyl group				

- In each case indicate how the following molecule can be made by drawing the reaction, naming all products and reactants, and naming the type of reaction that is occurring.
 - a) 1,2-dichlorocyclopentane (from a hydrocarbon)
 - b) 1,1-dichlorocyclopentane (from a hydrocarbon)
 - A condensation reaction to make butyl methanoate
 - d) The reactants from c) (not using hydrogenation)
- 9. Using Br₂(aq), how can you easily distinguish between pentane, 1-pentene, and 1-pentyne?
- 10. What is wrong with the following names?
 - a) 3-methyl-2-butyne
 - b) 1,2-dichlorocyclobutane
 - c) 3-methyloctene

- d) 2,3-dimethyl-4-ethylnonane
- e) 1-butanone
- f) 2-ethyl-2-methylhexane

Additional practice (pg. 1049-1053): 24.5, 7, 11, 12, 13, 14, 17, 19, 32, 36, 37

Modern Chemistry Review Carbon And Hydrocarbons Answer

G Psacharopoulos

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Table of Contents Modern Chemistry Review Carbon And Hydrocarbons Answer

- 1. Understanding the eBook Modern Chemistry Review Carbon And Hydrocarbons Answer
 - The Rise of Digital Reading Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Review Carbon And Hydrocarbons Answer
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Personalized Recommendations
 - Modern Chemistry Review Carbon And Hydrocarbons Answer User Reviews and Ratings
 - Modern Chemistry Review Carbon And Hydrocarbons Answer and Bestseller Lists

- 5. Accessing Modern Chemistry Review Carbon And Hydrocarbons Answer Free and Paid eBooks
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 - Modern Chemistry Review Carbon And Hydrocarbons Answer Budget-Friendly Options
- 6. Navigating Modern Chemistry Review Carbon And Hydrocarbons Answer eBook Formats
 - o ePub, PDF, MOBI, and More
 - Modern Chemistry Review Carbon And Hydrocarbons Answer Compatibility with Devices
 - Modern Chemistry Review Carbon And Hydrocarbons Answer Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Highlighting and Note-Taking Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Interactive Elements Modern Chemistry Review Carbon And Hydrocarbons Answer
- 8. Staying Engaged with Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Review Carbon And Hydrocarbons Answer
- 9. Balancing eBooks and Physical Books Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Modern Chemistry Review Carbon And Hydrocarbons Answer
- 10. Overcoming Reading Challenges
 - $\circ\,$ Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Setting Reading Goals Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Fact-Checking eBook Content of Modern Chemistry Review Carbon And Hydrocarbons Answer
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
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4

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