chemical Reactions and Equations

with new properties are formed.

The burning of magnesium in air to form magnesium acide

18 an example of a Chemical Reaction

magnesium + acygen Heat Magnesium oxide

(As ribbon) (From air) (White powden)

Note: The magnesium ribbon which we use usually has a coating of 'magnesium occide' on its swiface which is formed by the slow action of oxygen of air on it. So before burning in air, the magnesium ribbon is cleaned by rubbing with a sand paper. This is done to remove the protective layer of magnesium occide from the swiface of magnesium ribbon so that it may readily combine with the oxygen of air

Characteristics of Chemical Reactions

- (1) Evolution of a gas
- (11) Formation of a precipitate
- (m) change in colour
- (IV) change in temperature
- (v) change in state
- 1.) Evolution of a gas

The chemical reaction blu zinc and dilute sulphunic acid is characterised by the evolution of Hydrogen gas

2n + H2SO4 --- > 20504 + H2

Modern Chemistry Chemical Equations And Reactions Notes

Terry C. Jones

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