

Chemistry Exam - Gases

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Part 1 - Concepts and Vocabulary. Choose the best answer to each of the following questions. 3 pts each.

1. All of the following apply to gases EXCEPT

- a. Compressible
- b. Particles vibrate around fixed point
- c. Fluid
- d. Large amounts of space between particles

2. Which of the following is NOT an STP value?

- a. 101.3 kPa
- b. 1.00 atm
- c. 100 °C
- d. 160 Torr

3. The graph below is representative of ____ Law.



- a. Boyle's
- b. Charles'
- c. Avogadro's
- d. Gay-Lussac's

4. The graph below is representative of ____ Law.



- a. Boyle's
- b. Charles'
- c. Avogadro's
- d. Gay-Lussac's

5. The graph below is representative of ____ Law.



- a. Boyle's
- b. Charles'
- c. Avogadro's
- d. Gay-Lussac's

6. At STP, one mole of any gas will occupy ____ liters.

- a. 160
- b. 101.3
- c. 22.0
- d. 22.4

7. Which of the following gas relationships are inverse relationships?

- a. Pressure and temperature
- b. Pressure and volume
- c. Temperature and volume
- d. A & c

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KJ Lindholm-Leary



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