

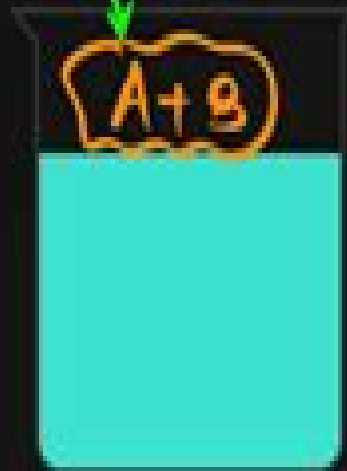


Molarity on dilution

$$M = \frac{n_B}{V_{\text{sol}}^n(\text{L})}$$

$$[n_B = M V]$$

(A)



B → no change

A → Volume ↑

moles before dilution = moles after diluting

$$M_1 V_1 = M_2 V_2$$

M_1 = molarity before dilution

V_1 = volume

M_2 = molarity after dilution

V_2 = volume

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Jessica J Manson



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