

Moz

# Molarity, Molality, & Solutions Practice

Molarity  $M = \frac{\text{moles}}{L}$

- Find the M of 1 mole of salt in 2.5 L. of water.

$$\frac{1 \text{ mole}}{2.5 L} = \boxed{.4 M}$$

- Find the M of 2.3 moles of salt in 1.14L. of water.

$$\frac{2.3 \text{ moles}}{1.14} = \boxed{2 M}$$

- Find the M of 1.9 moles of sodium acetate in 12.4L. of water.

$$\frac{1.9}{12.4} = \boxed{.15 M}$$

- How many moles of potassium carbonate are needed to make 500 mL. of a 1.25 M solution?

$$1.25 M = \frac{\text{moles}}{.5 L} \quad (.5)(1.25) = \text{moles} \quad \boxed{0.625 \text{ moles}}$$

- How many liters of 2.5 M solution can be made using 1.8 moles of lithium bromide?

$$2.5 M = \frac{1.8 \text{ moles}}{V} \quad \frac{(2.5)(V)}{2.5} = \frac{1.8}{2.5} \quad \boxed{0.72 L}$$

- What is the concentration of a 450 mL. solution that contains 200 grams of iron (II) chloride?

$$\frac{200 \text{ g FeCl}_2}{112.15 \text{ g}} \left| \frac{1 \text{ mole FeCl}_2}{112.15 \text{ g}} \right| = 1.77 \text{ moles} \quad \frac{1.77}{.45 L} = \boxed{3.93 M}$$

- How many grams of ammonium sulfate are needed to make 0.75L. of solution at a concentration of 4.0 M?

$$4 M = \frac{\text{moles}}{0.75} = 3 \text{ moles} \quad \frac{3 \text{ moles } (NH_4)_2SO_4}{132.17 \text{ g}} \left| \frac{1 \text{ mole } (NH_4)_2SO_4}{132.17 \text{ g}} \right| = \boxed{396.5 \text{ g}}$$

## Molality

- Calculate the molality of a solution containing 2.89g of NaCl dissolved in 0.159 L. of water.

$$\frac{2.89 \text{ g NaCl}}{58.44 \text{ g}} \left| \frac{1 \text{ mole}}{58.44 \text{ g}} \right| = 0.049 \text{ mol} \quad \frac{0.049}{0.159} = \boxed{.31 m}$$

- Calculate the molality of a solution containing 1.80 mol KCl in 16.0 mol of H<sub>2</sub>O

$$\frac{16 \text{ mol H}_2\text{O}}{1 \text{ mol}} \left| \frac{18 \text{ g}}{1 \text{ mol}} \right| = 288.32 \text{ g} = \text{mol} \quad \frac{1.8 \text{ mol}}{288 \text{ g}} = \boxed{6.25 m}$$

# Molarity Webquest Answer Key Teacher

**Rachel Sandford**



## **Molarity Webquest Answer Key Teacher:**

## Decoding **Molarity Webquest Answer Key Teacher**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Molarity Webquest Answer Key Teacher**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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web symphony no 4 op 29 fs 76 also known as the inextinguishable danish det uudslukkelige was completed by danish composer carl nielsen in 1916 composed against the backdrop of the first world war this symphony is among the most dramatic that nielsen wrote featuring a battle between two sets of timpani

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web the often raging fourth symphony finished in 1916 might be interpreted as a reaction to world war i nielsen however did not suggest this specifically but pointed to a universal conception when he said music is life and like life inextinguishable the composer s title no more than a philosophical guideline to the symphony is

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