

Name: _____ Date: _____

The Mole Webquest

Use the following website to answer this set of questions:

<http://antoinet.freefrang.co.uk/chem/sources/101/moles/faq/why-use-moles.shtml>

- 1) A mole of anything is how many? (give the number):
- 2) Why is it that different amounts of things can still equal one mole? (think about the weight of a dozen elephants vs a dozen eggs)
- 3) Why do we want to use the concept of moles?
- 4) Once we know the number of moles we can convert to the number of: _____ of _____ and vice versa.
- 5) How many grams of water are in one mole of water?
- 6) How many molecules of water are in one mole of water?

Use the following website to answer the next set of questions:

<https://chemfacts.wordpress.com/2015/02/06/mole-calculations-1/>

- 7) Define molar mass.
- 8) What is the mass of one mole of oxygen?
- 9) Find the molar mass of NH_4NO_3 . (Show your work)

Converting between grams, moles, and particles

- 10) Copy the chart:

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11) If you are converting between particles and moles, use _____ as conversion factor.

Mole Webquest Key

Thomas Elliott Taylor



Mole Webquest Key:

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