

Key

Molecular Geometry

How can molecular shapes be predicted using the VSEPR theory?

Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are *three-dimensional*. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

Model 1 – Lewis Structures

Lewis Structures		3-D Molecular Shape
1. H_2CO	3 electron domains (3 bonding, 0 nonbonding)	trigonal planar
2. BeF_2	2 electron domains (2 bonding, 0 nonbonding)	Linear
3. CH_4	4 electron domains (4 bonding, 0 nonbonding)	Tetrahedral
4. NH_3	4 electron domains (3 bonding, 1 nonbonding)	pyramidal
5. H_2O	4 electron domains (2 bonding, 2 nonbonding)	Bent
6. CO_2	2 electron domains (2 bonding, 0 nonbonding)	linear

Lone pair = ..

Molecular Geometry Pogil Activities For High School

J. C. D. Brand, J. C. Speakman



Molecular Geometry Pogil Activities For High School:

Molecular Geometry Alison Rodger, Mark Rodger, 2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms. The book is comprised of seven chapters that tackle several areas of molecular geometry. Chapter 1 reviews the definition and determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical changes. Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements beyond the second row. The book also talks about the complexes of transition metals and f block elements and then covers the organometallic compounds and transition metal clusters. The last chapter tackles the consequences of small local variations in geometry. The text will be of great use to chemists who primarily deal with the properties of molecules and atoms. The VSEPR Model of Molecular Geometry Ronald J Gillespie, Istvan Hargittai, 2013-03-21 Valence Shell Electron Pair Repulsion VSEPR theory is a simple technique for predicting the geometry of atomic centers in small molecules and molecular ions. This authoritative reference was written by Istvan Hargittai and the developer of VSEPR theory Ronald J Gillespie. In addition to its value as a text for courses in molecular geometry and chemistry, it constitutes a classic reference for professionals. Starting with coverage of the broader aspects of VSEPR, this volume narrows its focus to a succinct survey of the methods of structural determination. Additional topics include the applications of the VSEPR model and its theoretical basis. Helpful data on molecular geometries, bond lengths, and bond angles appear in tables and other graphics. The VSEPR Model of Molecular Geometry Ronald James Gillespie, István Hargittai, 1991-01 **Molecular Geometry** Ronald James Gillespie, 1972 Physical Methods for Determining Molecular Geometry Wallace S. Brey, 2000 *A Programmed Approach to Molecular Geometry* George I. Sackheim, *The Teaching of Molecular Geometry* Vincenzo P. Civitella, 1976 *Molecular geometry Second Edition* Gerardus Blokdyk, 2018 Molecular geometry Second Edition Molecular Geometry Optimization Jenn-Tai Hwang, 1973 *Physical Methods for Determining Molecular Geometry* Wallace Siegfried Brey (ifj), 1965 Lewis Structures and Molecular Geometry Jeffrey R. Buell, 1992 **Shape in Chemistry** Paul G. Mezey, 1993 **Rodger Molecular Geometry** Butterworth-Heinemann Limited, 1995-01-01 **The Determination of Molecular Structure** P. J. Wheatley, 1965 Determination of Molecular Geometry by Spectroscopic Methods, 1972 Molecular Structure J. C. D. Brand, J. C. Speakman, 1960 **Misconceptions in the Area of Molecular Geometry and Molecular Polarity Among College Introductory and General Chemistry Students** Patricia L. Rambo, 2004 **Methods for Improved Molecular Geometry and Frequency Calculations** Frank J. Dobek, 2017 *Lewis Structures and Molecular Geometry (computer File)*. Jeffrey R. Buell, 1968 **Molecular Geometry & Bonding** Clyde Metz, 1998-03-01

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