

11. (a) Write the recording techniques used in method study. Explain with a neat sketch of material type flow process chart. (16)

Or

- (b) What do you understand by Analytical estimating? Give the procedure for conducting work measurement study by analytical estimating. (16)

12. (a) Explain the use of computers in process planning and cost estimation and list out the advantages of CAPP. (16)

Or

- (b) What are factors influencing process selection and write down the process selection parameters. (16)

13. (a) (i) Write the difference between cost accounting and cost estimating. (8)
(ii) Write basic steps in cost estimation. (8)

Or

- (b) Calculate prime cost, factory cost, production cost, total cost and selling price per item from the data given below for the year 2012-13.

Cost of raw material in stock as on 01.04.2012	- Rs 25,000
Raw material purchased	- Rs 40,000
Direct labour cost	- Rs 14,000
Direct expenses	- Rs 1,000
Factory/work over heads	- Rs 9,750
Administrative expenditure	- Rs 6,500
Selling and distribution expenses	- Rs 3,250
Number of items produced	- 650
Cost of raw material in stock as on 31.03.2013	- Rs 15,000
Net Profit of the items is 10 percent of the total cost of the product.	

14. (a) (i) What are the three methods used in conceptual Cost estimation? Explain any two methods briefly. (8)
(ii) Write the data requirements and sources of information for cost estimation. (8)

Or

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 mL H_2SO_4 4 mL in the 125 mL flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 mL erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 Å bond distances this will result in an optimization failure as the atoms are even closer for the programme to accommodate their electrons demonstration of creating the p arenium cation by the nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 mL of nitric acid and 4.0 mL of sulfuric acid and will then react with 4.5 mL of bromobenzene this will produce o-bromonitrobenzene

and p bromonitrobenzene

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