

Chapter 9 Reaction rates

Time permitted: 30 minutes

	Section	Number of questions	Marks available
A	Multiple choice	15	15
B	Short answer	5	15
	Total		30

Scale:

A+	29–30	A	26–28	B	23–25	C	19–22	D	15–18	E	9–14	UG	0–8
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Section A Multiple choice (15 marks)

Section A consists of 15 questions, each worth one mark. Each question has only one correct answer. Circle the correct answer. Attempt all questions. Marks will not be deducted for incorrect answers. You are advised to spend no more than 15 minutes on this section.

- Why does reaction rate increase as temperature increases?
 - ☒ Particles move faster and this increases the probability of collision.
 - ☐ There are more particles at higher temperatures so more of them collide.
 - ☐ The activation energy is lowered.
 - ☐ Particles require a set temperature in which to begin colliding.

For questions 2 and 3 refer to the chemical reaction below.



- Which does not occur during the reaction?
 - ☒ The concentration of H_2O increases.
 - ☐ The mass of CaCO_3 decreases.
 - ☐ The volume of CO_2 increases.
 - ☐ The concentration of Ca^{2+} increases.
- The rate of the reaction could not be determined by:
 - ☐ recording the change in mass of the CaCO_3 over time.
 - ☐ recording the mass of CO_2 lost over time.
 - ☒ measuring the final mass of CaCO_3 .
 - ☐ collecting and recording the volume of CO_2 gas produced over time.

Lesson 36 Reaction Rates Answers Key

Minjie Lin



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Solution Processing of Inorganic Materials David Mitzi, 2008-12-22 Discover the materials set to revolutionize the electronics industry The search for electronic materials that can be cheaply solution processed into films while simultaneously providing quality device characteristics represents a major challenge for materials scientists Continuous semiconducting thin films with large carrier mobilities are particularly desirable for high speed microelectronic applications potentially providing new opportunities for the development of low cost large area flexible computing devices displays sensors and solar cells To date the majority of solution processing research has focused on molecular and polymeric organic films In contrast this book reviews recent achievements in the search for solution processed inorganic semiconductors and other critical electronic components These components offer the potential for better performance and more robust thermal and mechanical stability than comparable organic based systems Solution Processing of Inorganic Materials covers everything from the more traditional fields of sol gel processing and chemical bath deposition to the cutting edge use of nanomaterials in thin film deposition In particular the book focuses on materials and techniques that are compatible with high throughput low cost and low temperature deposition processes such as spin coating dip coating printing and stamping Throughout the text illustrations and examples of applications are provided to help the reader fully appreciate the concepts and opportunities involved in this exciting field In addition to presenting the state of the art research the book offers extensive background material As a result any researcher involved or interested in electronic device fabrication can turn to this book to become fully versed in the solution processed inorganic materials that are set to revolutionize the electronics industry **19 Years JEE MAIN Chapter-wise Solved Papers (2002 - 20) 12th Edition** Disha Experts, 2020-08-18

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