

# Chemistry 2 Student Workbook

## Answers to Worksheets

Teacher Notes

### UNIT 3

#### Worksheet 01 Crossword—exploring energy

##### ACROSS

- 3 Coal seam  
5 Enthalpy  
7 Exothermic  
11 Cathode  
12 Renewable  
14 Thermochemical  
18 Salt bridge  
19 Secondary  
21 Hygroscopic  
22 Fuel  
23 Anode

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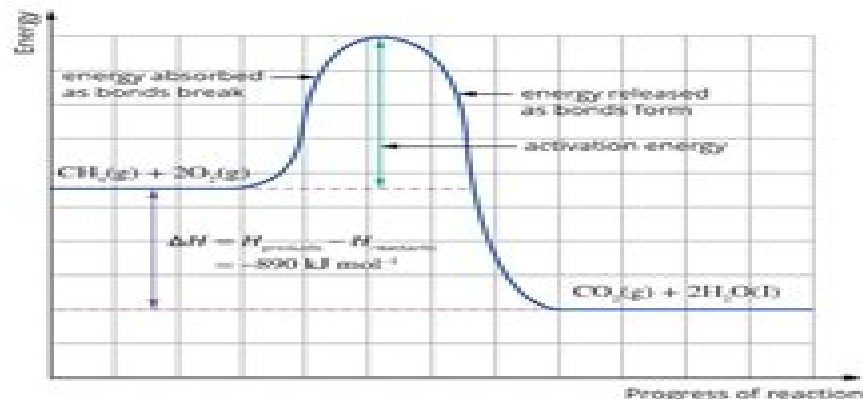
- 1 Galvanic  
2 Biofuel  
4 Spontaneous  
6 Biogas  
8 Joule  
9 Oxidation  
10 Greenhouse  
13 Activation  
15 Electrolyte  
16 Viscosity  
17 Reduction  
20 Fossil  
22 Fuel

#### Worksheet 02 Combustion of fuels—thermochemical equations and specific heat capacity

- 1 The combustion of fuels releases a significant quantity of energy that can be used as an **energy source**. **Chemical energy** is the energy stored in the bonds between atoms and between molecules arising from attractions and repulsions between subatomic particles. It is called **enthalpy** and is represented by the symbol, ***H***. All combustion reactions are **exothermic** reactions. In such reactions, the enthalpy change, symbol  **$\Delta H$** , has a **negative** sign.

$$\Delta H = H_{\text{products}} - H_{\text{reactants}}$$

2



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**Guillaume Favre**



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