# SNC2DI - REVIEW FOR CHEMISTRY

- Review "Key Terms" from Chapters 4, 5, & 6 (p. 139, 193, 221)
- Review "Unit Summary" listing key concepts, chapter summaries, vocabulary, and key visuals on pg 244
- Try "Chapter Review" questions:
  - Ch. 4 → pg 190-191 #4-6, 8, 11, 12-17, 19-20
  - Ch. 5 → pg 218-219 #1-3, 9-11, 13
  - Ch. 6 → pg 242-243 #1, 3, 6-8, 10-12, 14-20
- The following "Unit Review" questions:
  - pg 248-249 #2-11, 18-20, 25-27, #30-32
  - o pg 250-251 #43-44, 49-54, 59
  - pg 252-253 #60-61
- Study your notes/quizzes/labs from the beginning of the unit. Getting them organized first will make this much easier to do!!
- Make study notes to organize the information.

#### TOPIC LIST

- Periodic table
- Bohr diagrams of atoms & ions
- lons and the periodic table
- Chemical families names on the periodic table and their charges
- lonic compounds naming & formulas
- Polyatomic compounds naming & formulas
- Molecular compounds naming and formulas
- Types of chemical reactions (five)
- Law of conservation of mass
- Counting atoms
- Balancing equations
- Acids & bases
- pH scale

If you have any questions, please ask! I am here to help. Please come in at lunch or set up an appointment to meet before or after school if you would like additional help outside class time.

#### Good luck and happy studying!

# Chemistry Unit Review: ANSWERS

Unit Review Questions from Text: p.330-332 # 1a-f, 3 a-g,4-7, 10, 11, 12, 13, 14, 15, 16, 18

#### Atoms

 Fill in the blanks: "Nucleus with positive <u>protons</u> and neutral <u>neutrons</u>. Negative <u>electrons</u> in orbits outside nucleus. The electrons in the outermost orbit are called <u>valence electrons</u>, and they are involved in chemical reactions. If electrons are lost, then a <u>positive ion</u> forms, and if electrons are gained, then a <u>negative ion</u> forms."

# **Mcps Chemistry Exam Review**

Neil D. Jespersen, Pamela Kerrigan

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