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### CHEMISTRY PAPER-II

#### (MODEL PAPER)

Annual Examination 2021

Total Time: 2 hours.

Total Marks: 85

Time :30 min **SECTION 'A' (M.C.Qs. (Multiple Choice Questions) Marks: 43**

Note: This section consists of 43 questions. Attempt all M.C.Qs. Each carries 1 marks.

Q 1:- Choose the correct answers for each from the given options:

The element which is present in group V A and period 3<sup>rd</sup>, its atomic number is:

- \* 15                      \* 7                      \* 8                      \* 18

2) Hydrides of group V A are ----- in nature:

- \* Acidic                      \* Basic                      \* Amphoteric                      \* Neutral

3) Potassium when combines with oxygen form:

- \* Normal oxide                      \* Super oxide                      \* Per oxide                      \* All

4) Lithium has many similarities to its diagonal neighbour-----in "Be" family:

- \* K                      \* Mg                      \* Ca                      \* Na

5)  $\text{Li}^+_{\text{aq}}/\text{Li}$  couple has exceptionally high negative electrode potential because of its large value of

- \* Ionization potential                      \* Hydration enthalpy  
\* Electron Affinity                      \* Electronegativity

6) Aluminium does not react with Nitric acid due to

- \* Low reactivity                      \* High Ionization Potential  
\* Formation of oxide layer                      \* It is a metal

7) "A" is the element of III A group which "B" belongs to V A group. When "A" reacts with "B" forms:

- \*  $\text{A}_2\text{B}_3$                       \* AB                      \*  $\text{AB}_3$                       \*  $\text{A}_3\text{B}_2$

8) The mixture of "Al" and  $\text{Fe}_2\text{O}_3$  is used in :

- \* Pyrolysis                      \* Thermite process                      \* Electrolysis                      \* Washing

9) This element is solid at room temperature and pressure:

- \* Oxygen                      \* Fluorine                      \* Bromine                      \* Iodine

10) Electronic configuration of  $\text{Cu}^{+1}$ :

- \*  $4s^1, 3d^{10}$                       \*  $4s^0, 3d^9$                       \*  $4s^0, 3d^{10}$                       \*  $4s^2, 3d^{10}$

11) The coordination number of Pt in  $[\text{Pt}(\text{en})_2\text{Cl}_2]^{+2}$  is:

# Karachi Board Chemistry For Class Xii

**Sir Stanley Reed**



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## **Table of Contents Karachi Board Chemistry For Class Xii**

1. Understanding the eBook Karachi Board Chemistry For Class Xii
  - The Rise of Digital Reading Karachi Board Chemistry For Class Xii
  - Advantages of eBooks Over Traditional Books
2. Identifying Karachi Board Chemistry For Class Xii
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Karachi Board Chemistry For Class Xii
  - User-Friendly Interface
4. Exploring eBook Recommendations from Karachi Board Chemistry For Class Xii
  - Personalized Recommendations
  - Karachi Board Chemistry For Class Xii User Reviews and Ratings
  - Karachi Board Chemistry For Class Xii and Bestseller Lists
5. Accessing Karachi Board Chemistry For Class Xii Free and Paid eBooks
  - Karachi Board Chemistry For Class Xii Public Domain eBooks
  - Karachi Board Chemistry For Class Xii eBook Subscription Services
  - Karachi Board Chemistry For Class Xii Budget-Friendly Options
6. Navigating Karachi Board Chemistry For Class Xii eBook Formats

- ePub, PDF, MOBI, and More
- Karachi Board Chemistry For Class Xii Compatibility with Devices
- Karachi Board Chemistry For Class Xii Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Karachi Board Chemistry For Class Xii
  - Highlighting and Note-Taking Karachi Board Chemistry For Class Xii
  - Interactive Elements Karachi Board Chemistry For Class Xii
- 8. Staying Engaged with Karachi Board Chemistry For Class Xii
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Karachi Board Chemistry For Class Xii
- 9. Balancing eBooks and Physical Books Karachi Board Chemistry For Class Xii
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Karachi Board Chemistry For Class Xii
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Karachi Board Chemistry For Class Xii
  - Setting Reading Goals Karachi Board Chemistry For Class Xii
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Karachi Board Chemistry For Class Xii
  - Fact-Checking eBook Content of Karachi Board Chemistry For Class Xii
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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