

How to Balance:



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N₂ H₂ N_h3 Chemistry If8766

Justin L. Crossland



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Iron Coordination Chemistry of N₂, N₂H₂, N₂H₄, and NH₃ Justin L. Crossland, 2009 **Solubilities of NH₃, CO, CO₂, and SF₆** Elfreda T. Chang, Nev A. Gokcen, Talmadge M. Poston, AEROSPACE CORP EL SEGUNDO CALIF LAB OPERATIONS., 1970 The solubilities of gases at various temperatures and pressures in the following gas liquid binary systems were investigated Gaseous ammonia NH₃ in liquid hydrazine N₂H₄ in monomethylhydrazine MMH N₂H₃CH₃ and in unsymmetrical dimethylhydrazine UDMH 1 1 N₂ H₂ CH₃ 2 gaseous carbon monoxide CO in liquid UDMH gaseous sulfur hexafluoride SF₆ in liquid MMH and in dinitrogen tetroxide N₂O₄ and gaseous carbon dioxide CO₂ in N₂O₄ The solubilities of NH₃ were found to obey Henry's law in the range of about 0 to 0.20 mole fraction All other gases are much less soluble than NH₃ and obey Henry's law at all concentrations For each liquid the standard Gibbs energy of solution of each gas has been expressed as a function of absolute temperature Author

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