

Naming and Equations Review

Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method!

- 1) LiBr _____
- 2) Na_2O _____
- 3) CaSO_4 _____
- 4) $\text{Be}(\text{NO}_3)_2$ _____
- 5) NO _____

Write the formulas for the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right method!

- 6) sodium nitride _____
- 7) sulfur dioxide _____
- 8) ammonia _____
- 9) calcium phosphate _____
- 10) aluminum nitrate _____

Balance the following equations:

- 11) _____ $\text{NaCN} +$ _____ $\text{CuCO}_3 \rightarrow$ _____ $\text{Na}_2\text{CO}_3 +$ _____ $\text{Cu}(\text{CN})_2$
- 12) _____ $\text{Al}(\text{OH})_3 +$ _____ $\text{Cs} \rightarrow$ _____ $\text{CsOH} +$ _____ Al
- 13) _____ $\text{MgS} +$ _____ $\text{OF}_2 \rightarrow$ _____ $\text{MgF}_2 +$ _____ SO
- 14) _____ $\text{Se} +$ _____ $\text{H}_2 \rightarrow$ _____ H_2S
- 15) _____ $\text{LiCl} +$ _____ $\text{Br}_2 \rightarrow$ _____ $\text{LiBr} +$ _____ Cl_2
- 16) _____ $\text{Pb}(\text{NO}_3)_2 +$ _____ $\text{HF} \rightarrow$ _____ $\text{HNO}_3 +$ _____ PbF_2
- 17) _____ $\text{GaBr}_3 +$ _____ $\text{Na}_2\text{SO}_4 \rightarrow$ _____ $\text{NaBr} +$ _____ $\text{Ga}_2(\text{SO}_4)_3$
- 18) _____ $\text{P}_2\text{S}_2 \rightarrow$ _____ $\text{P}_4 +$ _____ S_8

Naming And Equations Review

Tobias Bleicker



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Table of Contents Naming And Equations Review

1. Understanding the eBook Naming And Equations Review
 - The Rise of Digital Reading Naming And Equations Review
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming And Equations Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming And Equations Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming And Equations Review
 - Personalized Recommendations
 - Naming And Equations Review User Reviews and Ratings
 - Naming And Equations Review and Bestseller Lists

5. Accessing Naming And Equations Review Free and Paid eBooks
 - Naming And Equations Review Public Domain eBooks
 - Naming And Equations Review eBook Subscription Services
 - Naming And Equations Review Budget-Friendly Options
6. Navigating Naming And Equations Review eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming And Equations Review Compatibility with Devices
 - Naming And Equations Review Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming And Equations Review
 - Highlighting and Note-Taking Naming And Equations Review
 - Interactive Elements Naming And Equations Review
8. Staying Engaged with Naming And Equations Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming And Equations Review
9. Balancing eBooks and Physical Books Naming And Equations Review
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming And Equations Review
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Naming And Equations Review
 - Setting Reading Goals Naming And Equations Review
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Naming And Equations Review
 - Fact-Checking eBook Content of Naming And Equations Review
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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